

Phonons

Thermal properties

1. Determine the dispersion relation:

Write down the equations of motion (masses and springs).

The solutions to these equations will be

$$A_{\mathbf{k}} \exp \left(i \left(\vec{k} \cdot \vec{a}_1 + \vec{k} \cdot \vec{a}_2 + \vec{k} \cdot \vec{a}_3 - \omega t \right) \right)$$

Substitute the solutions into the equations of motion to determine the dispersion relation.

2. Determine the density of states numerically from the dispersion relation

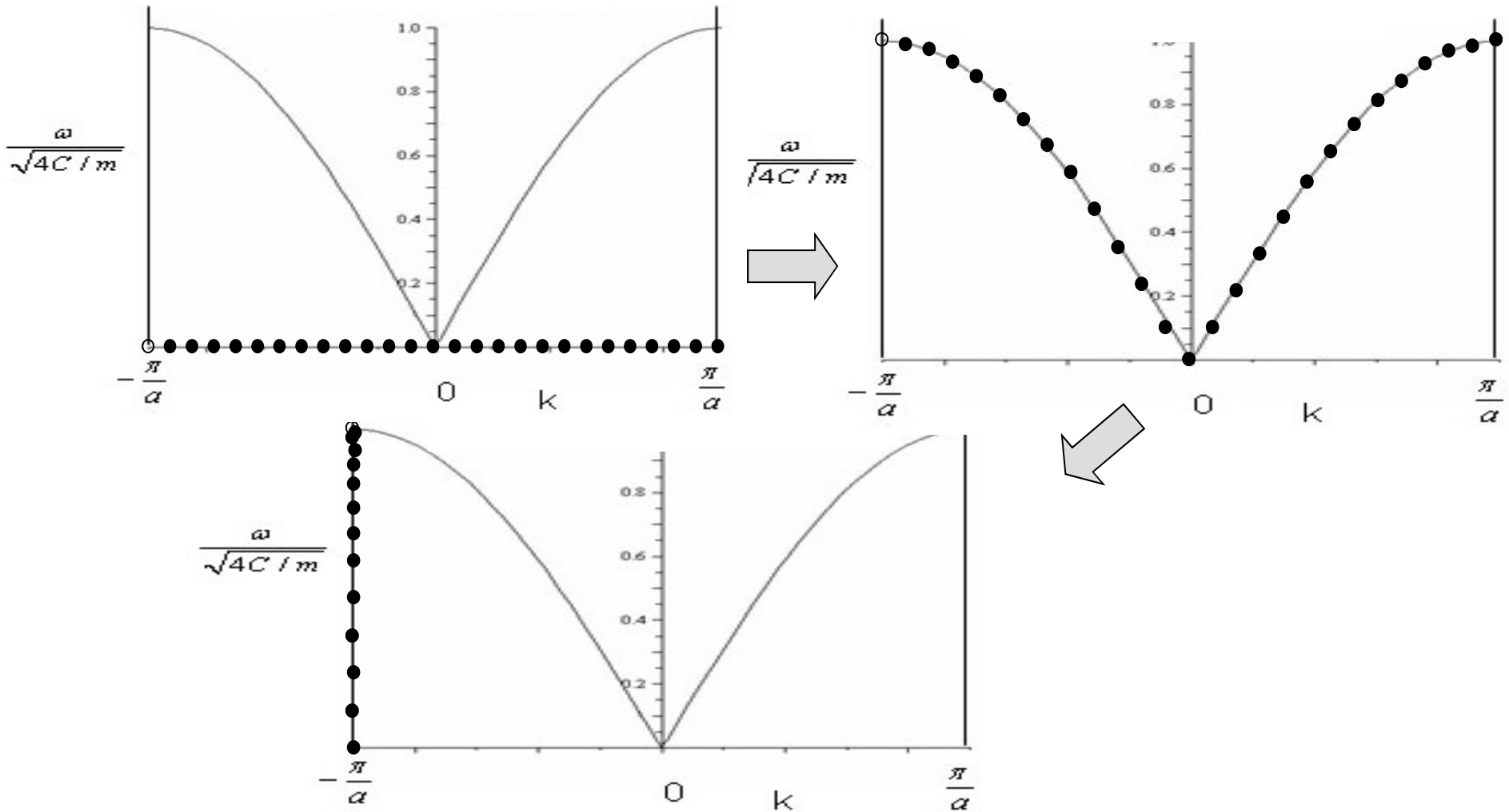
$$D(\omega)$$

For every allowed k , find all corresponding values of ω .

Linear Chain - density of states

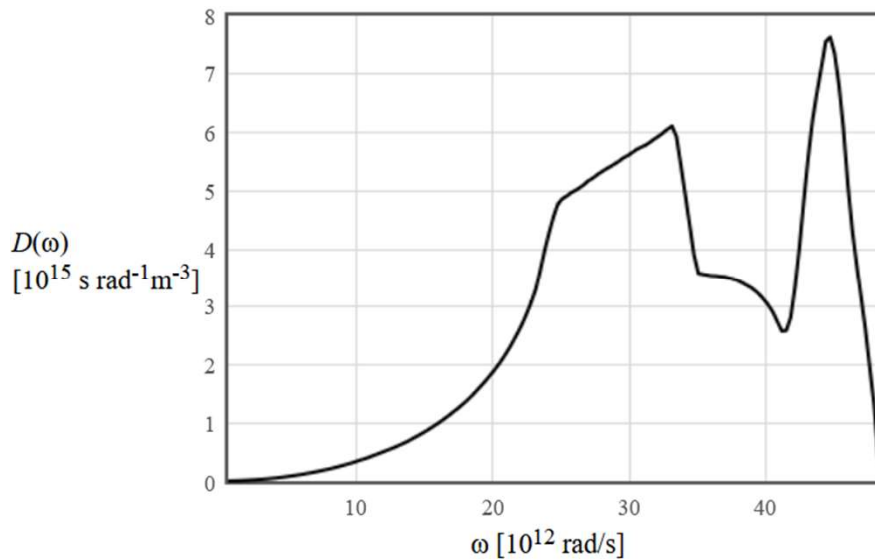
Determine the density of states numerically

$$\omega = \sqrt{\frac{4C}{m}} \left| \sin\left(\frac{ka}{2}\right) \right|$$

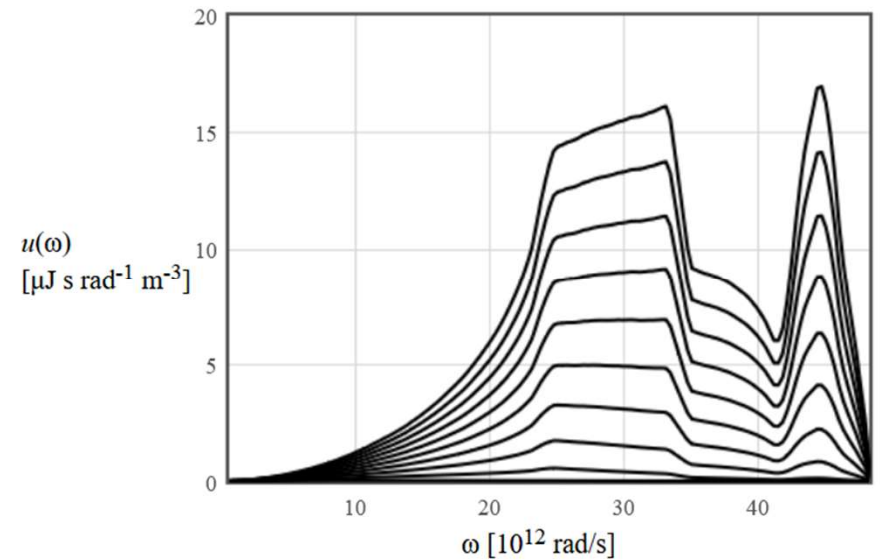


Density of states \rightarrow energy spectral density

$$u(\omega) = \frac{\hbar\omega D(\omega)}{\exp\left(\frac{\hbar\omega}{k_B T}\right) - 1}$$

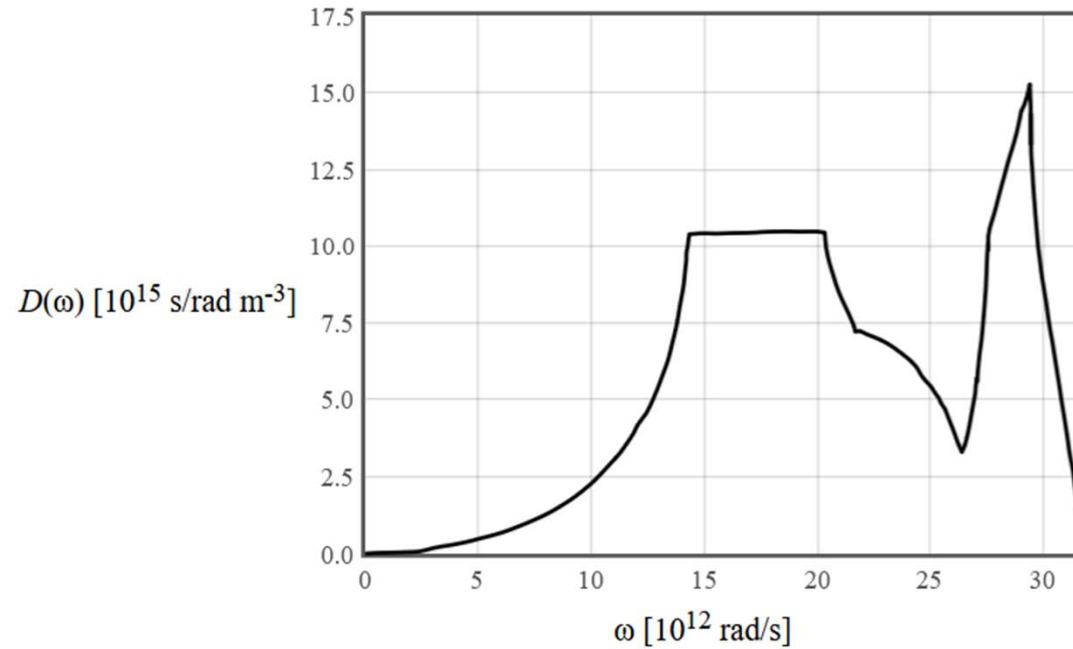


DoS \rightarrow $u(\omega)$



<http://lampx.tugraz.at/~hadley/ss1/phonons/table/dos2uw.html>

Phonon density of states for fcc silver



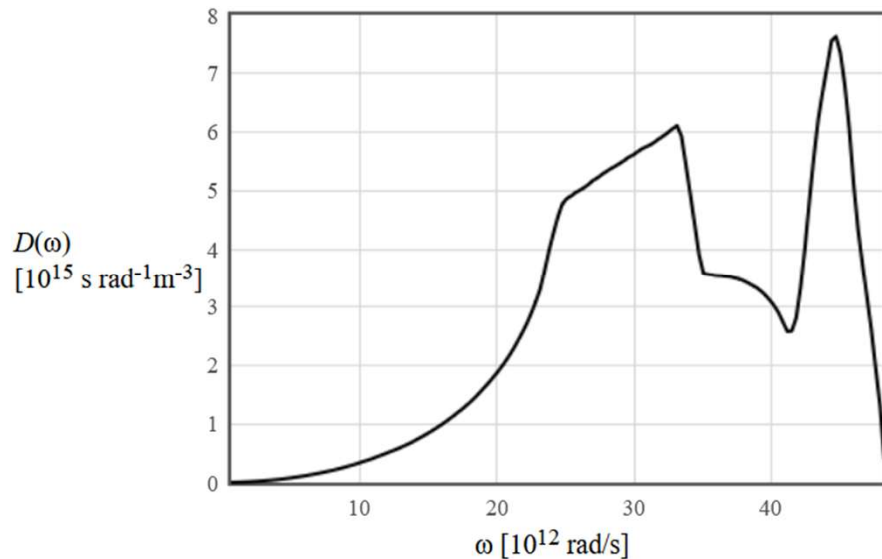
The atomic density is taken to be $5.86 \times 10^{28} \text{ m}^{-3}$. Each atom has three degrees of freedom so the integral over all frequencies is $3 \times 5.86 \times 10^{28} \text{ m}^{-3}$. The data is from [doi: 10.1007/b19988](https://doi.org/10.1007/b19988).

T = 296 K

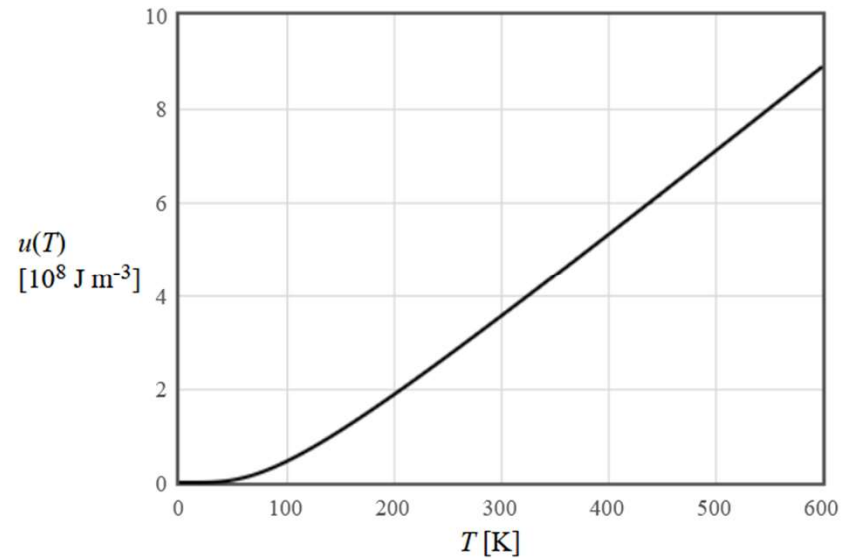
ω [rad/s]	$D(\omega)$ [s rad ⁻¹ m ⁻³]
0.0000	0.0000
5.7327e+10	6.8161e+12
4.0123e+11	2.3856e+13
7.4510e+11	3.0672e+13
1.0890e+12	3.4080e+13
1.4233e+12	4.0897e+13
1.7624e+12	5.1121e+13
2.0967e+12	5.7937e+13
2.4120e+12	7.4977e+13
2.7177e+12	1.2610e+14
3.0379e+12	1.8744e+14
3.3723e+12	2.3516e+14
3.7163e+12	2.7261e+14

Density of states \rightarrow Internal energy density

$$u(T) = \int_0^{\infty} \frac{\hbar\omega D(\omega)}{\exp\left(\frac{\hbar\omega}{k_B T}\right) - 1} d\omega$$



DoS \rightarrow $u(T)$



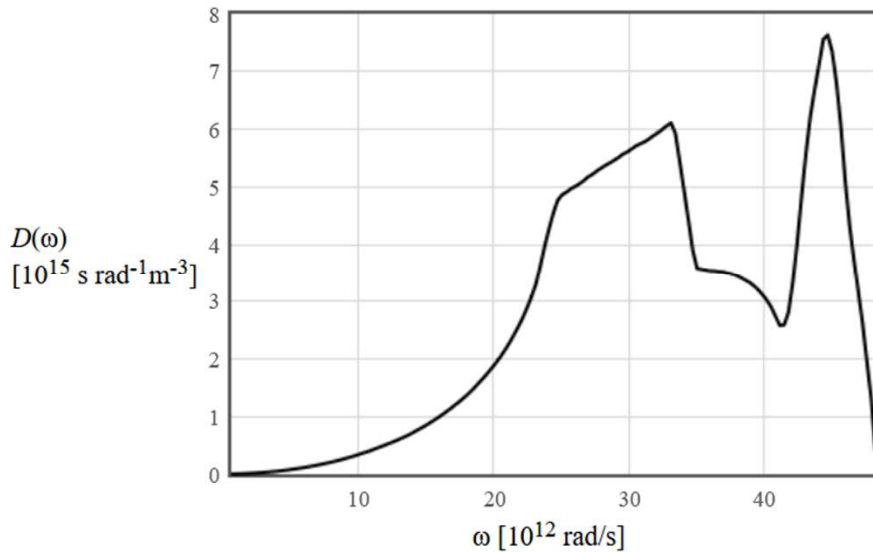
<http://lampx.tugraz.at/~hadley/ss1/phonons/table/dos2ut.html>

Specific Heat

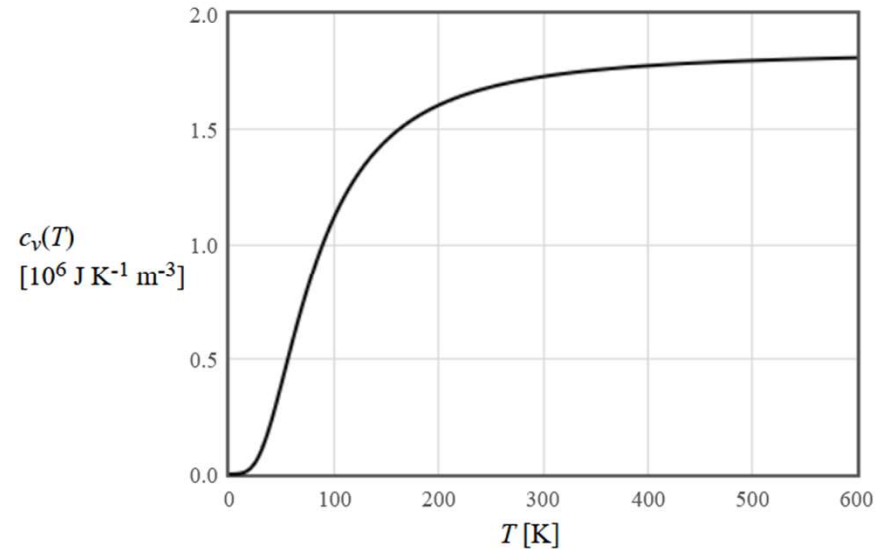
$$c_v = \left(\frac{\partial u}{\partial T} \right)_{N,V}$$

$$c_v = \int \hbar\omega D(\omega) \frac{\partial}{\partial T} \left(\frac{1}{e^{\frac{\hbar\omega}{k_B T}} - 1} \right) d\omega$$

$$c_v = \int \left(\frac{\hbar\omega}{T} \right)^2 \frac{D(\omega) e^{\frac{\hbar\omega}{k_B T}}}{k_B \left(e^{\frac{\hbar\omega}{k_B T}} - 1 \right)^2} d\omega$$



DoS \rightarrow cv(T)



<http://lampx.tugraz.at/~hadley/ss1/phonons/table/dos2cv.html>

Heat capacity / specific heat

Heat capacity is the measure of the heat energy required to increase the temperature of an object by a certain temperature interval.

Specific heat is the measure of the heat energy required to increase the temperature of a unit quantity of a substance by a certain temperature interval.

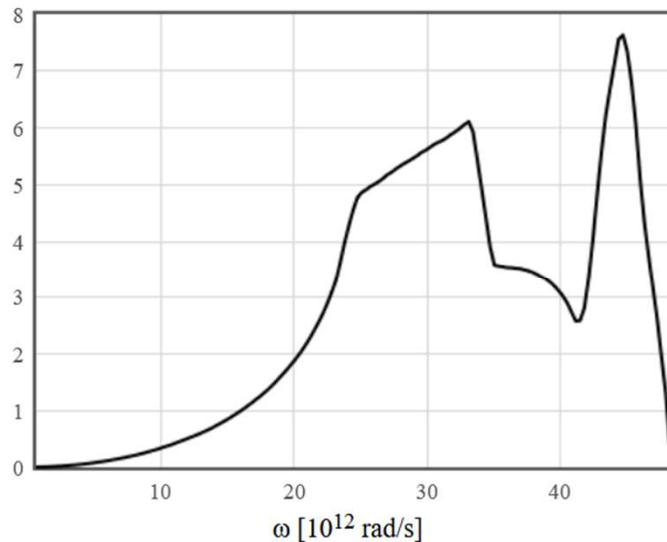
For solids, the heat capacity at constant volume and heat capacity at constant pressure are almost the same.

The heat capacity was historically important for understanding solids.

Density of states \rightarrow entropy density

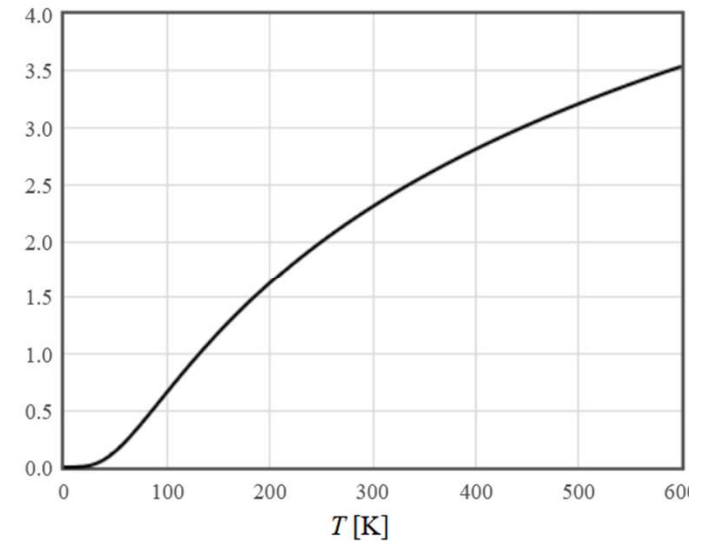
$$s = -\frac{\partial f}{\partial T} = -k_B \int_0^{\infty} D(\omega) \left(\ln \left(1 - e^{-\hbar\omega/k_B T} \right) + \frac{\hbar\omega}{k_B T \left(1 - e^{-\hbar\omega/k_B T} \right)} \right) d\omega$$

$D(\omega)$
[10^{15} s
 $\text{rad}^{-1}\text{m}^{-3}$]



DoS \rightarrow s(T)

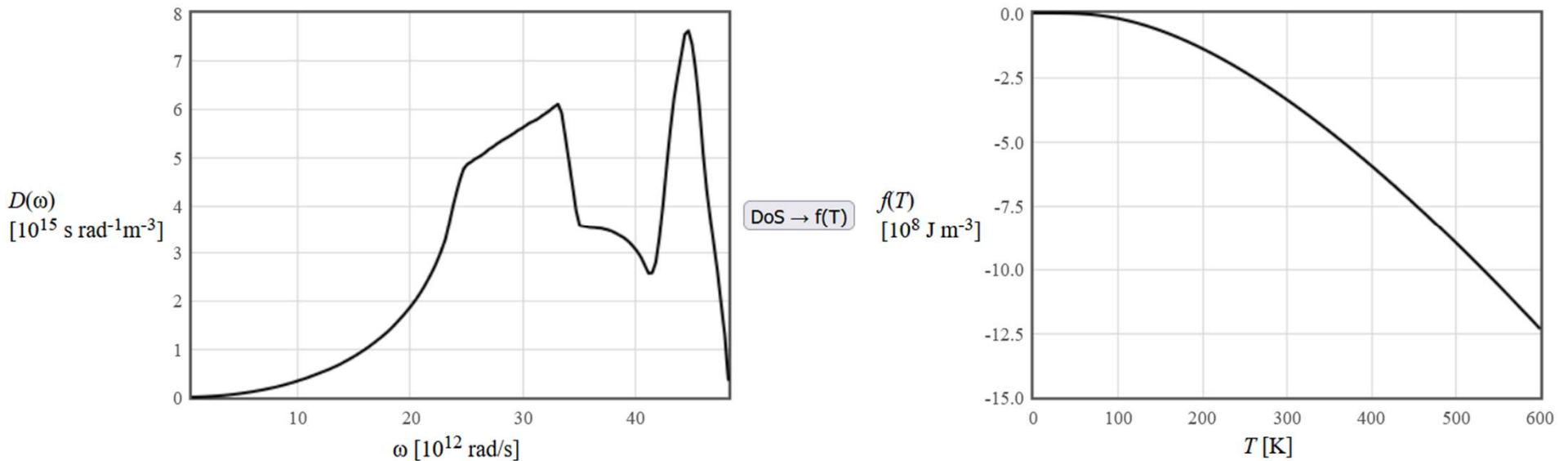
$s(T)$ [10⁶
 $\text{J K}^{-1} \text{m}^{-3}$]



<http://lampx.tugraz.at/~hadley/ss1/phonons/table/dos2s.html>

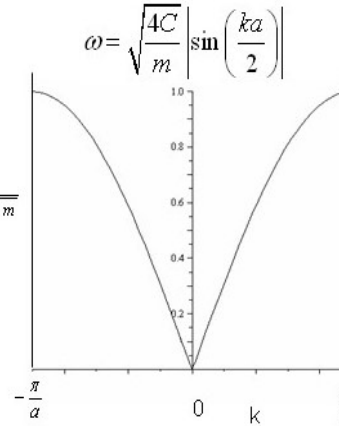
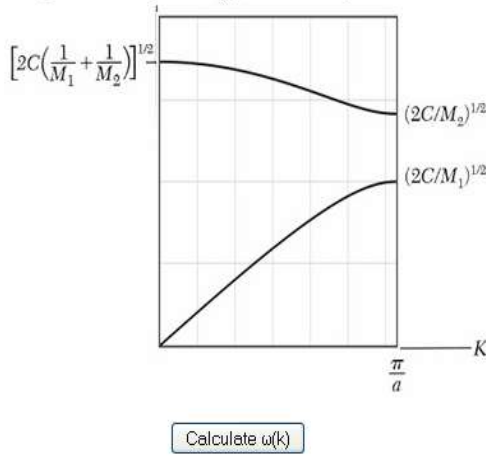
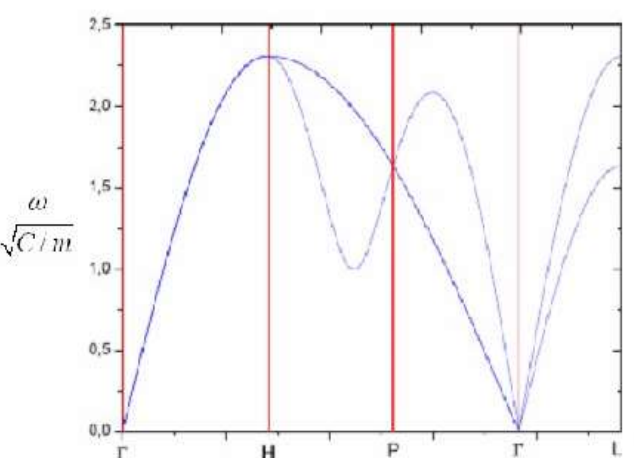
Density of states \rightarrow Helmholtz free energy density

$$f(T) = k_B T \int_0^{\infty} D(\omega) \ln \left(1 - \exp \left(\frac{-\hbar \omega}{k_B T} \right) \right) d\omega.$$



<http://lampx.tugraz.at/~hadley/ss1/phonons/table/dos2h.html>

Phonons

	<p style="text-align: center;">Linear Chain</p> $m \frac{d^2 u_s}{dt^2} = C(u_{s+1} - 2u_s + u_{s-1})$	<p style="text-align: center;">Linear chain 2 masses</p> $M_1 \frac{d^2 u_s}{dt^2} = C(v_{s-1} - 2u_s + v_s)$ $M_2 \frac{d^2 v_s}{dt^2} = C(u_s - 2v_s + u_{s+1})$	<p style="text-align: right;"><u>body centered cubic</u></p> $\frac{d^2 u_{lmn}^x}{dt^2} = \frac{C}{\sqrt{3} m} [(u_{l+1m+1n+1}^x - u_{lmn}^x) + (u_{l-1m+1n+1}^x - u_{lmn}^x) + (u_{l+1m-1n+1}^x - u_{lmn}^x) + (u_{l+1m+1n-1}^x - u_{lmn}^x) + (u_{l-1m+1n-1}^x - u_{lmn}^x) + (u_{l+1m+1n+1}^y - u_{lmn}^y) - (u_{l-1m+1n+1}^y - u_{lmn}^y) - (u_{l+1m-1n+1}^y - u_{lmn}^y) - (u_{l+1m+1n-1}^y - u_{lmn}^y) + (u_{l-1m+1n-1}^y - u_{lmn}^y) + (u_{l+1m-1n-1}^y - u_{lmn}^y) + (u_{l+1m+1n+1}^z - u_{lmn}^z) - (u_{l-1m+1n+1}^z - u_{lmn}^z) - (u_{l+1m-1n+1}^z - u_{lmn}^z) - (u_{l+1m+1n-1}^z - u_{lmn}^z) + (u_{l-1m+1n-1}^z - u_{lmn}^z) - (u_{l+1m-1n-1}^z - u_{lmn}^z)]$ <p style="text-align: right;">And similar expressions for the y and z</p>
<p>Eigenfunction solutions</p>	$u_s = A e^{i(ksa - \omega t)}$	$u_s = u e^{i(ksa - \omega t)}$ $v_s = v e^{i(ksa - \omega t)}$	$u_{lmn}^x = u \frac{x}{k} e^{i(l \vec{k} \cdot \vec{a}_1 + m \vec{k} \cdot \vec{a}_2 + n \vec{k} \cdot \vec{a}_3)} = u \frac{x}{k} e^{i(\dots)}$ <p style="text-align: right;">And similar expressions for the y and z</p>
<p>Dispersion relation</p>	$\omega = \sqrt{\frac{4C}{m}} \left \sin\left(\frac{ka}{2}\right) \right $ 	$\omega^2 = C \left(\frac{1}{M_1} + \frac{1}{M_2} \right) \pm C \sqrt{\left(\frac{1}{M_1} + \frac{1}{M_2} \right)^2 - \frac{4 \sin^2\left(\frac{ka}{2}\right)}{M_1 M_2}}$ 	<p style="text-align: right;">The dispersive</p> $\begin{matrix} 4 - \cos(\frac{\alpha}{2}(k_x + k_y + k_z)) - \cos(\frac{\alpha}{2}(3k_x - k_y - k_z)) & -\cos(\frac{\alpha}{2}(k_x + k_y - k_z)) \\ -\cos(\frac{\alpha}{2}(-k_x + 3k_y - k_z)) - \cos(\frac{\alpha}{2}(-k_x - k_y + 3k_z)) - \frac{m\omega^2}{\sqrt{3}C} & +\cos(\frac{\alpha}{2}(-k_x + 3k_y - k_z)) \\ -\cos(\frac{\alpha}{2}(k_x + k_y + k_z)) + \cos(\frac{\alpha}{2}(3k_x - k_y - k_z)) & 4 - \cos(\frac{\alpha}{2}(k_x + k_y - k_z)) \\ +\cos(\frac{\alpha}{2}(-k_x + 3k_y - k_z)) - \cos(\frac{\alpha}{2}(-k_x - k_y + 3k_z)) & -\cos(\frac{\alpha}{2}(-k_x + 3k_y - k_z)) \\ -\cos(\frac{\alpha}{2}(k_x + k_y + k_z)) + \cos(\frac{\alpha}{2}(3k_x - k_y - k_z)) & -\cos(\frac{\alpha}{2}(k_x + k_y - k_z)) \\ -\cos(\frac{\alpha}{2}(-k_x + 3k_y - k_z)) + \cos(\frac{\alpha}{2}(-k_x - k_y + 3k_z)) & +\cos(\frac{\alpha}{2}(-k_x + 3k_y - k_z)) \end{matrix}$ 
<p>Density of states $D(k)$</p>	$D(k) = \frac{1}{\pi}$	$D(k) = \frac{1}{\pi}$	$D(k) = \frac{3k^2}{2\pi^2}$

Thermal properties

internal energy density $u = \int_0^{\infty} u(\omega) d\omega = \int_0^{\infty} \frac{\hbar\omega D(\omega)}{\exp\left(\frac{\hbar\omega}{k_B T}\right) - 1} d\omega \quad [\text{J/m}^3]$

specific heat $c_v = \frac{du}{dT} = \int \left(\frac{\hbar\omega}{T}\right)^2 \frac{D(\omega) \exp\left(\frac{\hbar\omega}{k_B T}\right)}{k_B \left(\exp\left(\frac{\hbar\omega}{k_B T}\right) - 1\right)^2} d\omega \quad [\text{J K}^{-1} \text{ m}^{-3}]$

entropy density $s(T) = \int \frac{c_v}{T} dT = \frac{1}{T} \int_0^{\infty} \frac{\hbar\omega D(\omega)}{\exp\left(\frac{\hbar\omega}{k_B T}\right) - 1} d\omega \quad [\text{J K}^{-1} \text{ m}^{-3}]$

Helmholtz free energy density

$$f(T) = u - Ts = k_B T \int_0^{\infty} D(\omega) \ln \left(1 - \exp\left(\frac{-\hbar\omega}{k_B T}\right) \right) d\omega \quad [\text{J/m}^3]$$

Quartz

α -Quartz
trigonal
2.65 g/cm³

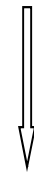
573°C
⇨

β -Quartz
hexagonal
2.53 g/cm³

870°C
⇨

β -Tridymite
hexagonal
2.25 g/cm³

1470°C



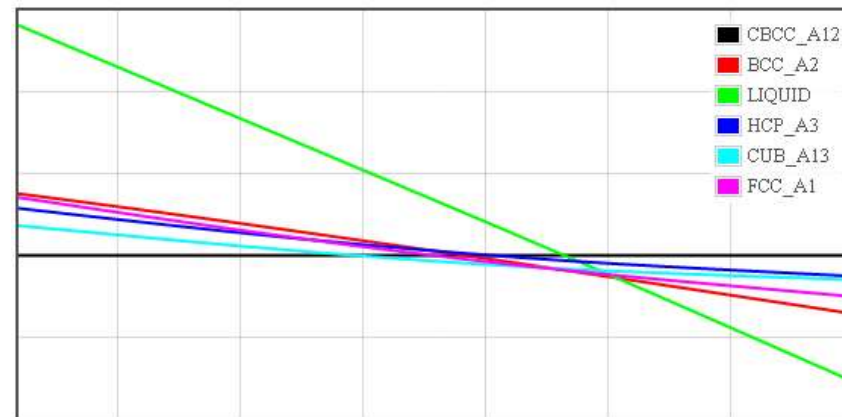
β -Cristobalite
cubic
2.20 g/cm³

Silica Melt

1705°C
⇨



$f-f_0$



T

GaPO₄

Material Constants

GaPO₄ (gallium phosphate) is a high temperature piezoelectric crystal, with superior sensing properties without any pyroelectric characteristics. This crystal has been used in technical applications such as pressure sensors since 1994.

GENERAL DATA

Point group	32 (D ₃) (quartz-homeotype)
Lattice constants (25°C)	a = 4.901 Å, c = 11.048 Å
Density (25°C)	3570 kg/m ³

PIEZOELECTRIC CONSTANTS

	25°C	500°C	700°C	950°C
d ₁₁ ^{a,b} [pC/N]	4.5	4.5	4.5	4.1 ^c
d ₁₄ ^a [pC/N]	1.9	1.6	1.4 ^c	1.0 ^c

$$d = \begin{pmatrix} d_{11} & -d_{11} & 0 & d_{14} & 0 & 0 \\ 0 & 0 & 0 & 0 & -d_{14} & -2d_{11} \\ 0 & 0 & 0 & 0 & 0 & 0 \end{pmatrix}$$

THERMAL EXPANSION COEFFICIENTS

at T₀ = 25°C, -253°C < T < 900°C

	α _{ii} (T ₀) [10 ⁻⁶ K ⁻¹]	Tα _{ii} ⁽¹⁾ [10 ⁻⁹ K ⁻²]	Tα _{ii} ⁽²⁾ [10 ⁻¹² K ⁻³]	Tα _{ii} ⁽³⁾ [10 ⁻¹⁵ K ⁻⁴]
α ₁₁	12.78	10.6	-16.1	12.3
α ₃₃	3.69	5.0	-5.4	3.6

$$\alpha = \begin{pmatrix} \alpha_{11} & 0 & 0 \\ 0 & \alpha_{11} & 0 \\ 0 & 0 & \alpha_{33} \end{pmatrix}$$

$$\alpha_{ii}(T) = \alpha_{ii}(T_0) + \sum_{n=1}^3 [T\alpha_{ii}^{(n)} \cdot (T - T_0)^n]$$

THERMAL CONDUCTIVITY

	50°C	70°C	100°C	130°C	150°C	180°C	200°C
λ ₁₁ [W m ⁻¹ K ⁻¹]	4.21	3.96	3.68	3.28	3.07	2.85	2.71
λ ₃₃ [W m ⁻¹ K ⁻¹]	6.66	6.14	5.66	5.00	4.78	4.27	4.02

$$\lambda = \begin{pmatrix} \lambda_{11} & 0 & 0 \\ 0 & \lambda_{11} & 0 \\ 0 & 0 & \lambda_{33} \end{pmatrix}$$

ELECTRIC RESISTIVITY

	25°C	200°C	300°C	500°C	700°C	900°C
$\rho[\Omega\text{m}]$	$> 10^{15}$	$> 10^{13}$	$> 10^{11}$	$> 10^9$	$> 10^7$	$> 10^5$

ELASTIC CONSTANTS at $T_0 = 25^\circ\text{C}$, $-50^\circ\text{C} < T < 700^\circ\text{C}$

	c_{ij} [GPa]	$Tc_{ij}^{(1)}$ [10^{-6}K^{-1}]	$Tc_{ij}^{(2)}$ [10^{-9}K^{-2}]	$Tc_{ij}^{(3)}$ [10^{-12}K^{-3}]
c_{11}^E	66.58	-44.1	-28.5	-59.4
c_{12}^E	21.81	-226.7	-70.8	-205.7
c_{13}^E	24.87	-57.6	41.3	-109.9
c_{14}^{Eb}	3.91	507.2	280.6	-99.9
c_{33}^E	102.13	-127.5	-18.3	-134.8
c_{44}^E	37.66	-0.4	-43.8	-37.1
c_{66}^E	22.38	44.9	-7.9	11.9

$$c = \begin{pmatrix} c_{11} & c_{12} & c_{13} & c_{14} & 0 & 0 \\ c_{12} & c_{11} & c_{13} & -c_{14} & 0 & 0 \\ c_{13} & c_{13} & c_{33} & 0 & 0 & 0 \\ c_{14} & -c_{14} & 0 & c_{44} & 0 & 0 \\ 0 & 0 & 0 & 0 & c_{44} & c_{14} \\ 0 & 0 & 0 & 0 & c_{14} & \frac{(c_{11} - c_{12})}{2} \end{pmatrix}$$

$$c_{ij}(T) = c_{ij}(T_0) \left[1 + \sum_{n=1}^3 Tc_{ij}^{(n)} \cdot (T - T_0)^n \right]$$

	s_{ij} [$10^{-12} \text{m}^2 \text{N}^{-1}$]	$Ts_{ij}^{(1)}$ [10^{-6}K^{-1}]	$Ts_{ij}^{(2)}$ [10^{-9}K^{-2}]	$Ts_{ij}^{(3)}$ [10^{-12}K^{-3}]
s_{11}^E	17.93	22.4	30.5	62.4
s_{12}^E	-4.82	-210.5	-0.1	-271.3
s_{13}^E	-3.19	181.6	78.2	322.2
s_{14}^{Eb}	-2.36	482.2	315.5	7.9
s_{33}^E	11.35	147.9	14.1	261.5
s_{44}^E	27.04	18.7	54.7	52.0
s_{66}^E	45.51	-26.9	24.0	-8.3

$$s = \begin{pmatrix} s_{11} & s_{12} & s_{13} & s_{14} & 0 & 0 \\ s_{12} & s_{11} & s_{13} & -s_{14} & 0 & 0 \\ s_{13} & s_{13} & s_{33} & 0 & 0 & 0 \\ s_{14} & -s_{14} & 0 & s_{44} & 0 & 0 \\ 0 & 0 & 0 & 0 & s_{44} & 2s_{14} \\ 0 & 0 & 0 & 0 & 2s_{14} & 2(s_{11} - s_{12}) \end{pmatrix}$$

$$s_{ij}(T) = s_{ij}(T_0) \left[1 + \sum_{n=1}^3 Ts_{ij}^{(n)} \cdot (T - T_0)^n \right]$$

RELATIVE DIELECTRIC CONSTANTS

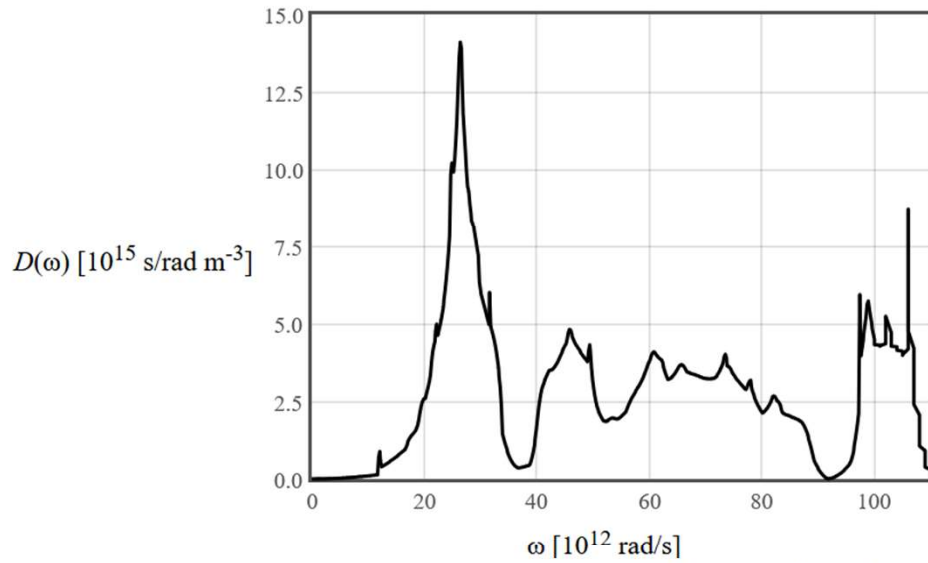
at $T_0 = 25^\circ\text{C}$, 1 kHz

	ϵ_{11}^T	ϵ_{11}^S	ϵ_{33}^S
	6.1	5.8	6.6
	6.6	6.6	

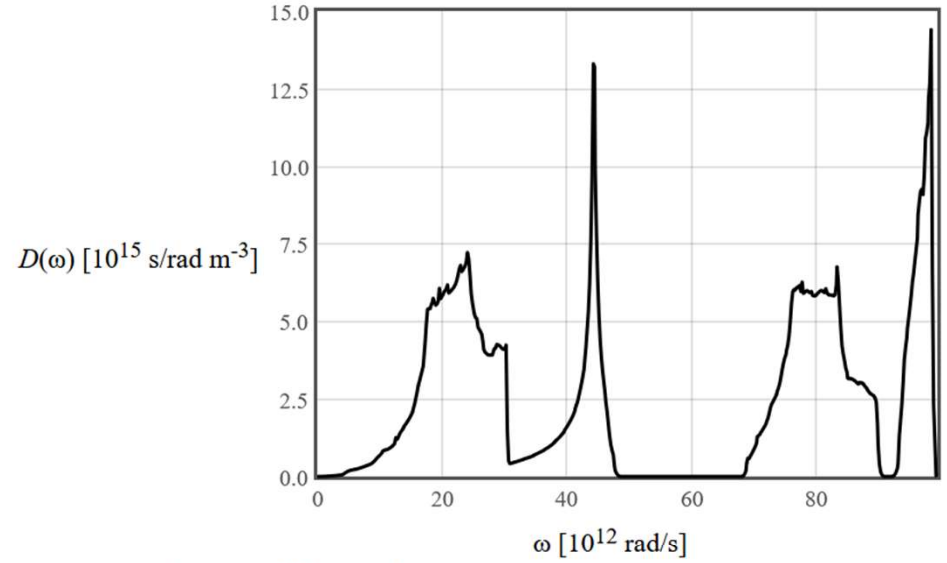
$$\epsilon = \begin{pmatrix} \epsilon_{11} & 0 & 0 \\ 0 & \epsilon_{11} & 0 \\ 0 & 0 & \epsilon_{33} \end{pmatrix}$$

ZnO

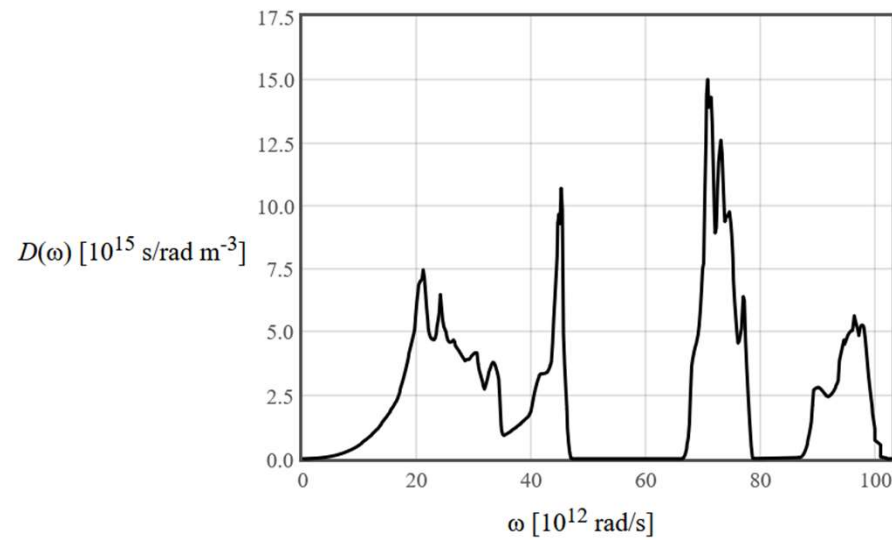
Phonon density of states for ZnO (Rocksalt)



Phonon density of states for ZnO (Zincblende)

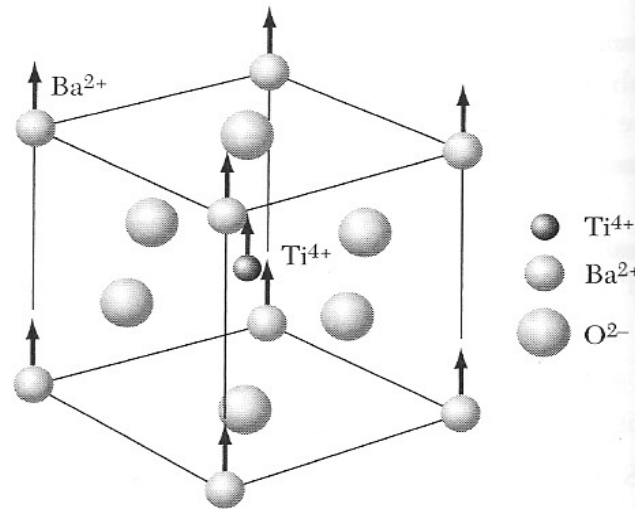


Phonon density of states for ZnO (Wurtzite)



Ferroelectricity

ABX₃
Perovskites

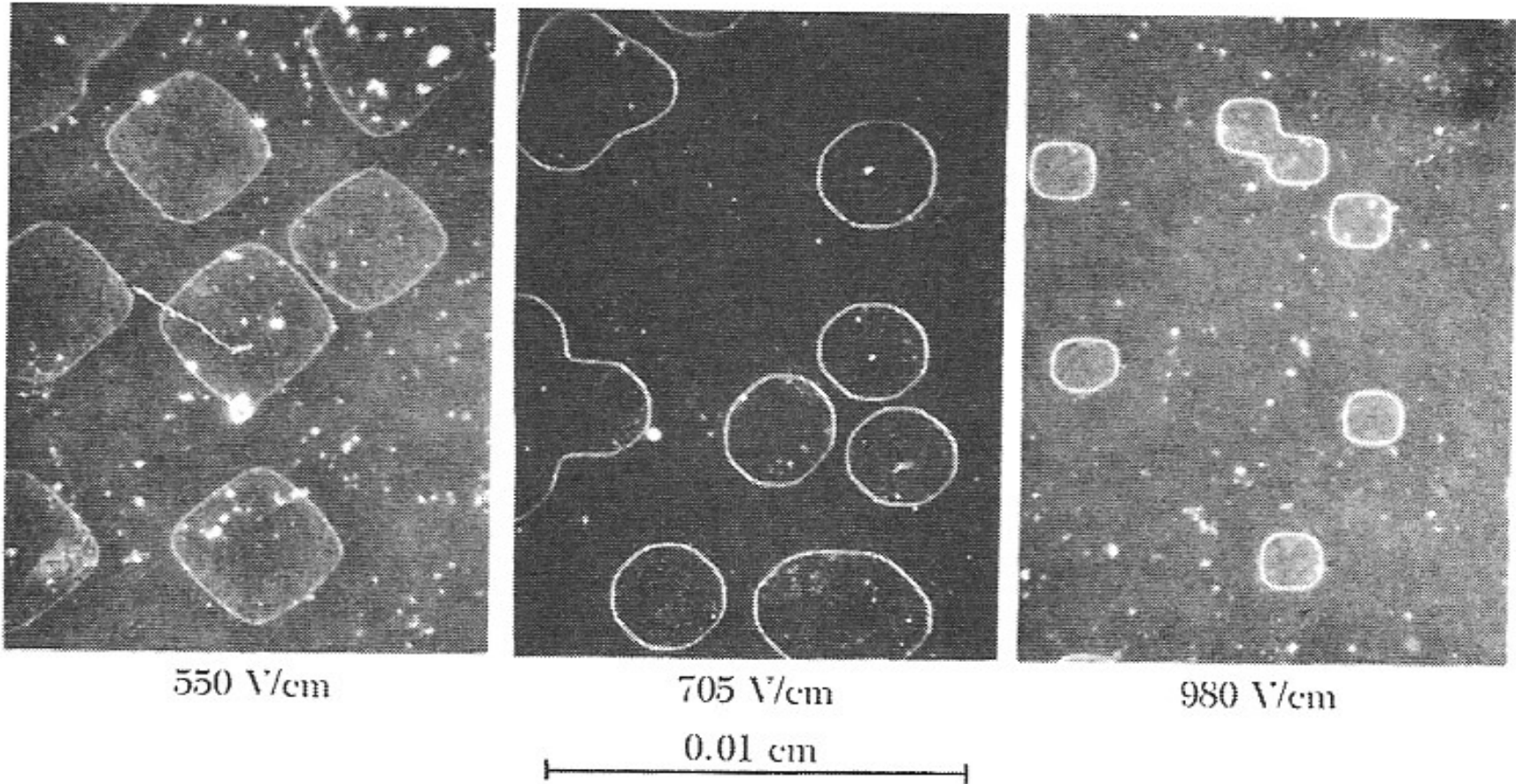


Spontaneous polarization
 Analogous to ferromagnetism
 Structural phase transition
 T_c is transition temperature

Electric field inside the material,
 is not conducting

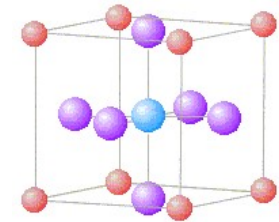
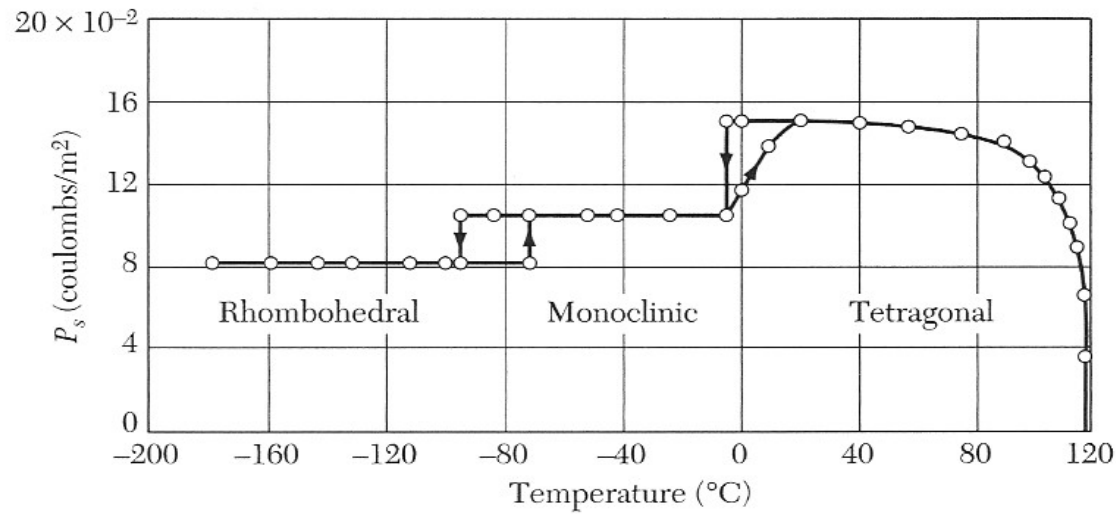
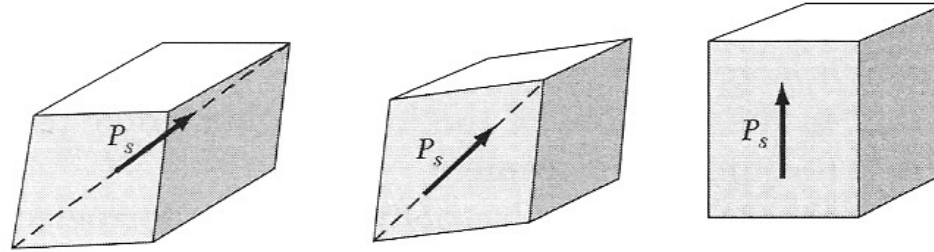
		T_c , in K	P_s , in $\mu\text{C cm}^{-2}$, at T K	
KDP type	KH ₂ PO ₄	123	4.75	[96]
	KD ₂ PO ₄	213	4.83	[180]
	RbH ₂ PO ₄	147	5.6	[90]
	KH ₂ AsO ₄	97	5.0	[78]
	GeTe	670	—	—
TGS type	Tri-glycine sulfate	322	2.8	[29]
	Tri-glycine selenate	295	3.2	[283]
Perovskites	BaTiO ₃	408	26.0	[296]
	KNbO ₃	708	30.0	[523]
	PbTiO ₃	765	>50	[296]
	LiTaO ₃	938	50	
	LiNbO ₃	1480	71	[296]

Ferroelectric domains



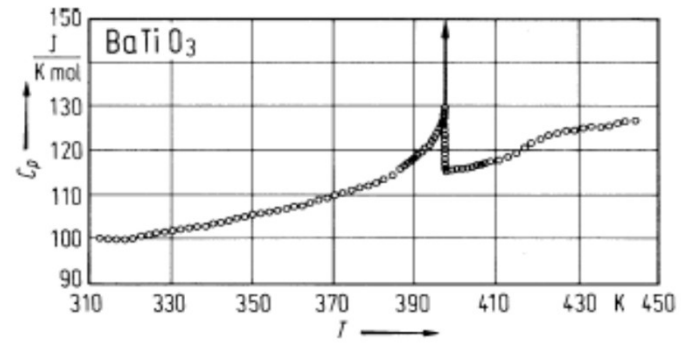
Increasing the electric field polarizes the material.

BaTiO₃



cubic (contains i = > no spontaneous P)

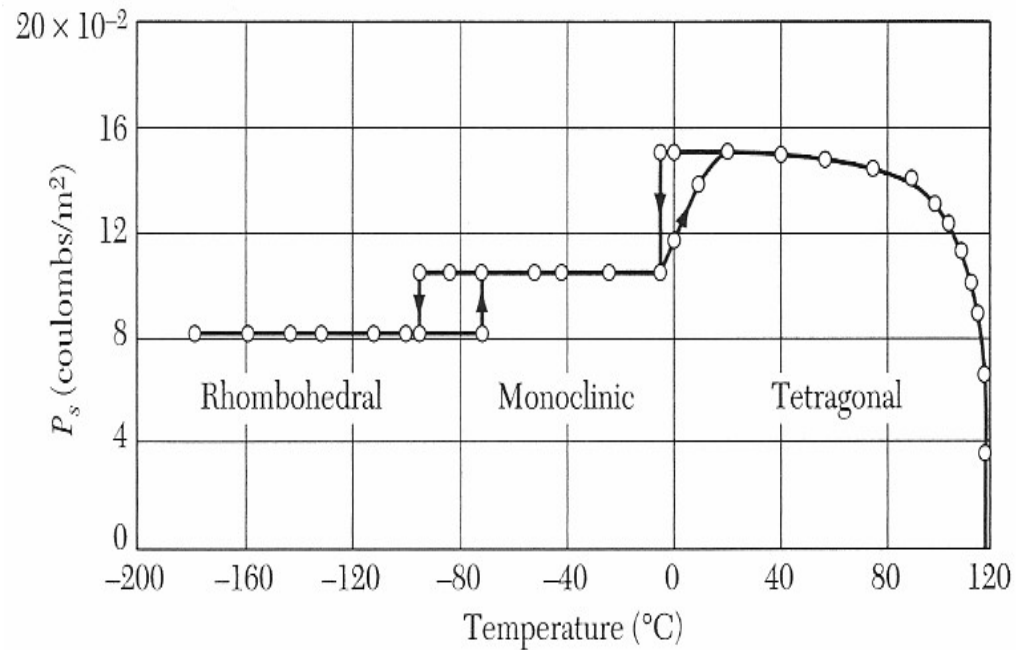
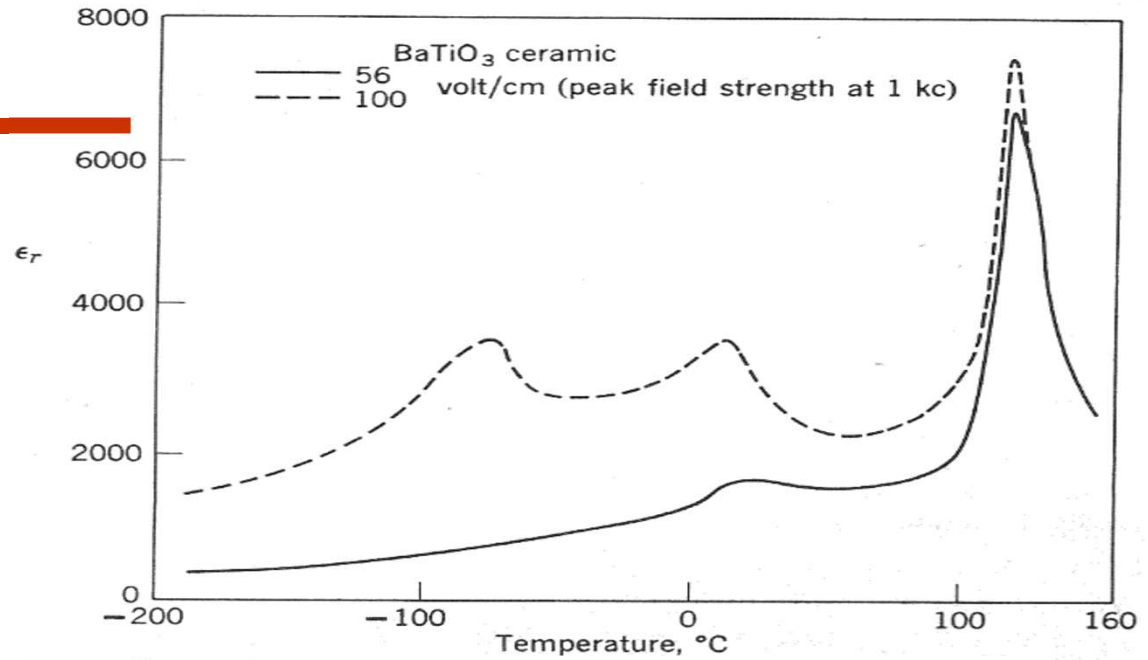
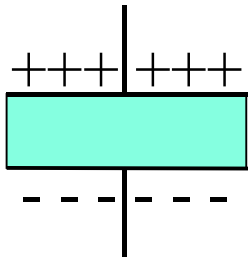
Can be used to make nonvolatile memory



BaTiO₃

$$\epsilon_r = \chi + 1$$

Can be used to make
ultracapacitors

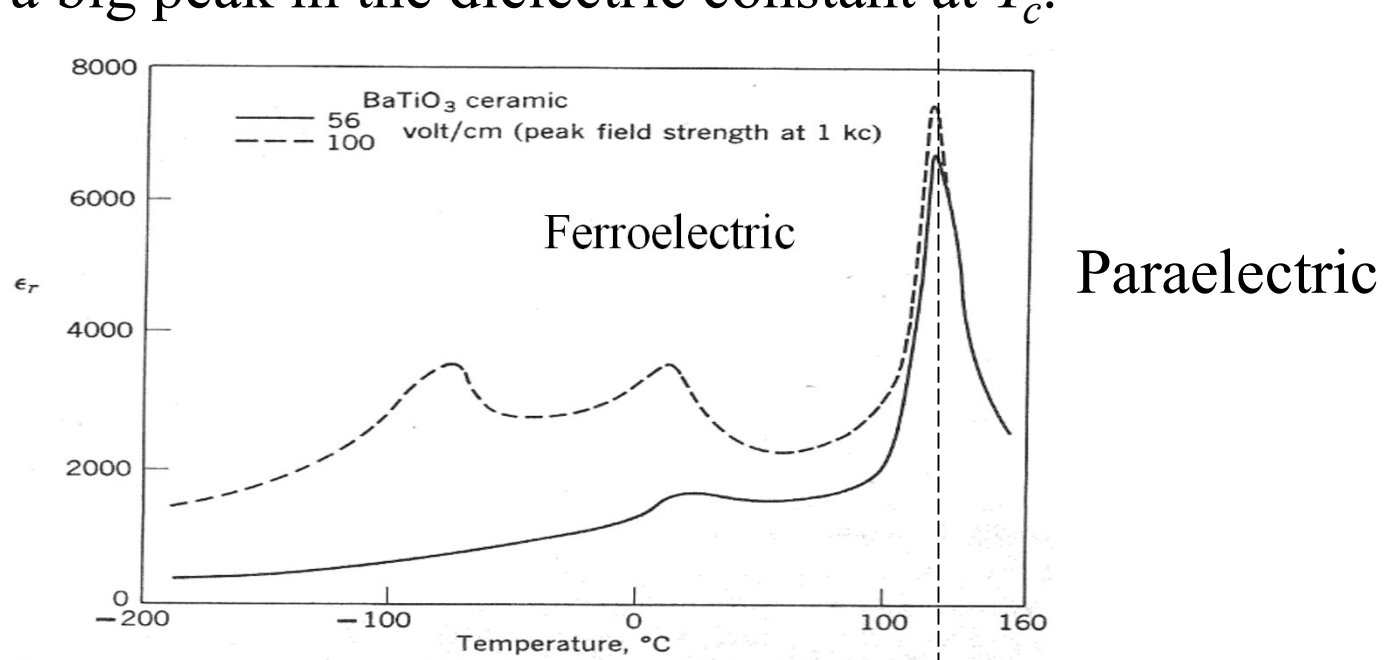


Paraelectric state

Above T_c , BaTiO₃ is paraelectric. The susceptibility (and dielectric constant) diverge like a Curie-Weiss law.

$$\chi \propto \frac{1}{T - T_c} \quad \epsilon = (1 + \chi) \epsilon_0$$

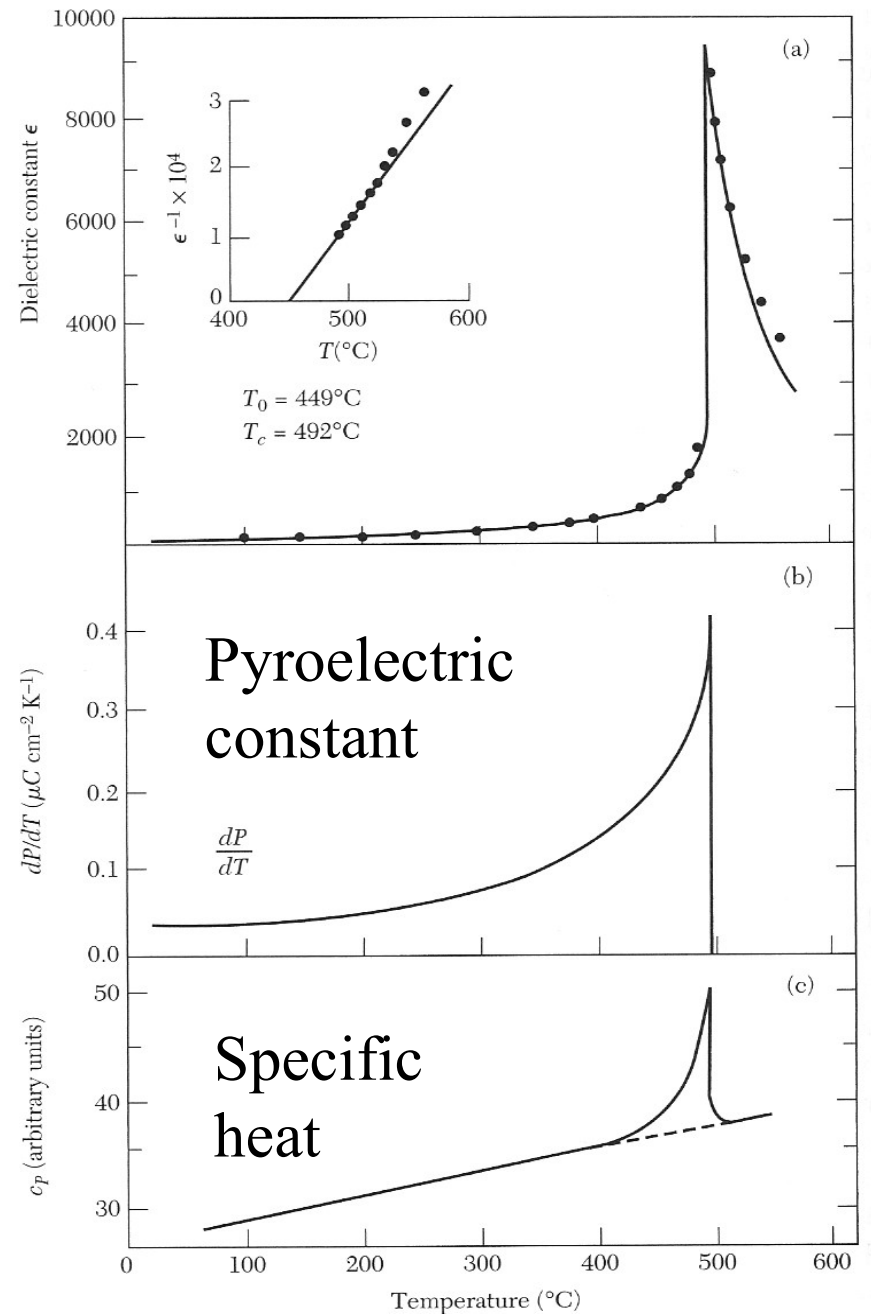
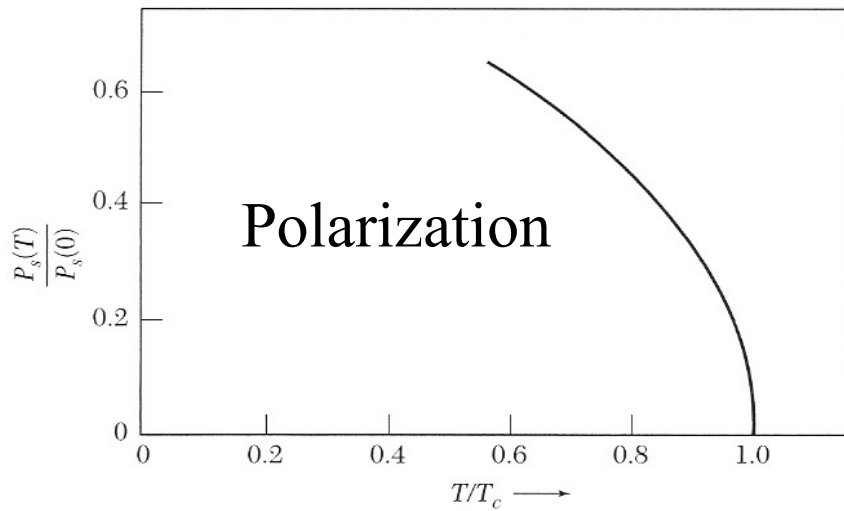
This causes a big peak in the dielectric constant at T_c .



PbTiO₃

Dielectric constant

$$\epsilon \propto \frac{1}{T - T_c}$$



Waves and particles

The eigen function solutions of the wave equation are plane waves. The scattering time is one over the rate for scattering from a given plane wave solution to any other.

Phonons are particles. The scattering time is the time before the phonons scatter and randomly change energy and momentum.

$$E = \hbar\omega$$

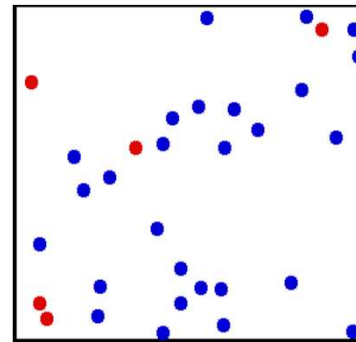
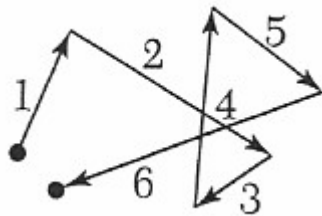
$$\vec{p} = \hbar\vec{k}$$

The average time between scattering events is $\tau_{sc} = 1/\Gamma$

Heat transport (Kinetic theory)

Treat phonons as an ideal gas of particles that are confined to the volume of the solid.

Phonons move at the speed of sound. They scatter due to imperfections in the lattice and anharmonic terms in the Hamiltonian.



The average time between scattering events is τ_{sc}

The average distance traveled between scattering events is the mean free path: $l = v\tau_{sc} \sim 10 \text{ nm}$

Diffusion equation/ heat equation

Diffusion constant $\frac{dn}{dt} = -D\nabla^2 n$

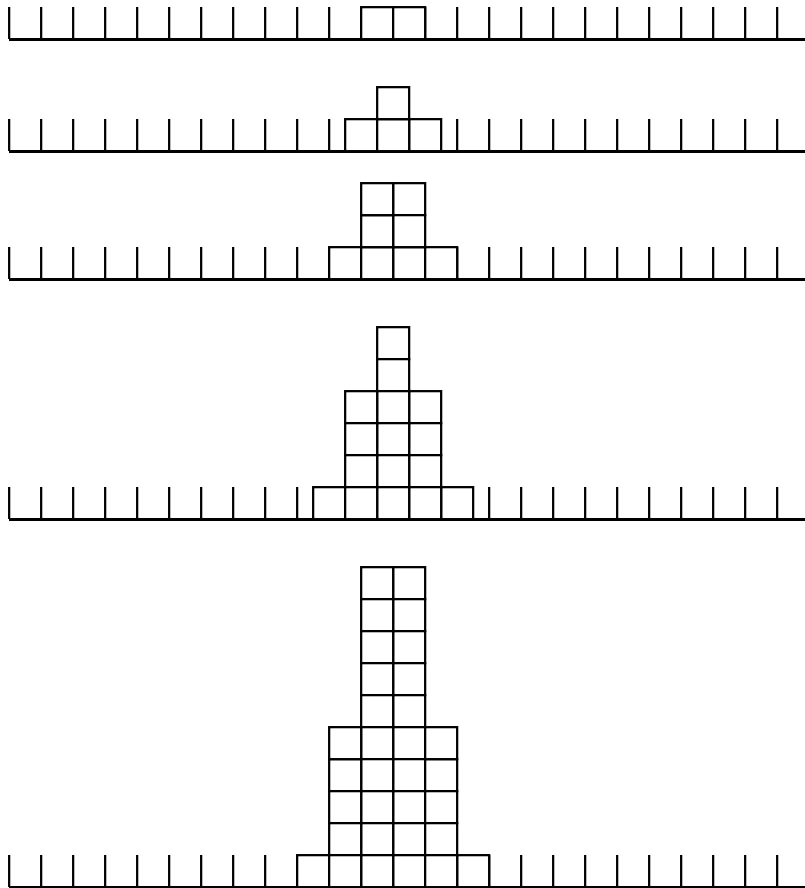
Fick's law $\vec{j} = -D\nabla n$

Continuity equation $\frac{dn}{dt} = \nabla \cdot \vec{j}$



$$n = \frac{1}{\sqrt{4\pi Dt}} \exp\left(\frac{-r^2}{4Dt}\right)$$

Random walk



$$\frac{\Delta n_s}{\Delta t} = n_{s+1} - 2n_s + n_{s-1}$$



$$\exp(-x^2)$$

Central limit theorem: A function convolved with itself many times forms a Gaussian

Thermal conductivity

$$\vec{j}_U = \bar{E} \vec{j}$$

Average particle energy

$$u = \bar{E} n$$

internal energy density

$$\vec{j}_U = -\bar{E} D \nabla n = -D \nabla u$$

$$\vec{j}_U = -D \frac{du}{dT} \nabla T = -D c_v \nabla T$$

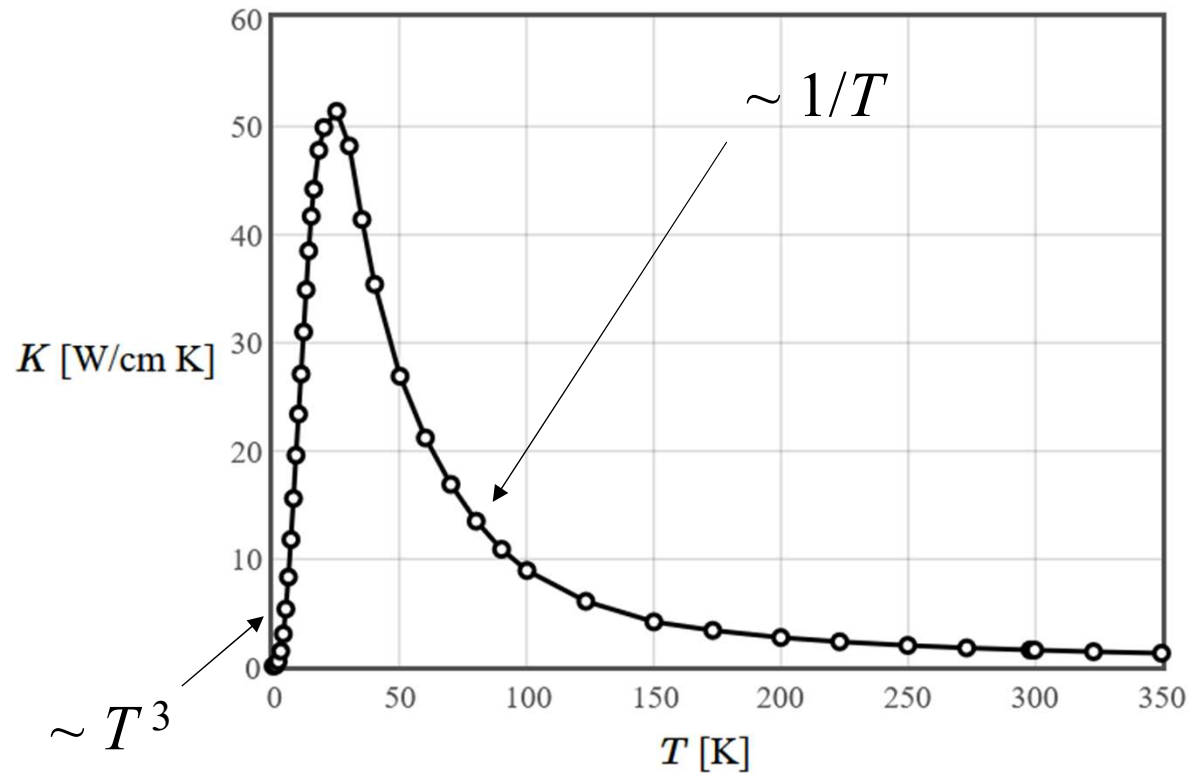
$$\vec{j}_U = -K \nabla T$$

Thermal conductivity

$$K = D c_v$$

$$K \rightarrow 0 \quad \text{as} \quad T \rightarrow 0$$

Thermal conductivity $\vec{j}_U = -K\nabla T$



The thermal conductivity of silicon.[1]

Thermal conductivity

$$\vec{j}_U = -K\nabla T$$

Material	Thermal conductivity W/(m·K)
Glass	1.1
Concrete, stone	1.7
Ice	2
Sandstone	2.4
Sapphire	35
Stainless steel	12.11 ~ 45.0
Lead	35.3
Aluminum	237
Aluminum alloys	120—180
Gold	318
Copper	401
Silver	429
Diamond	900 - 2320
Graphene	(4840±440) - (5300±480)

