

# Crystal physics

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# Strain

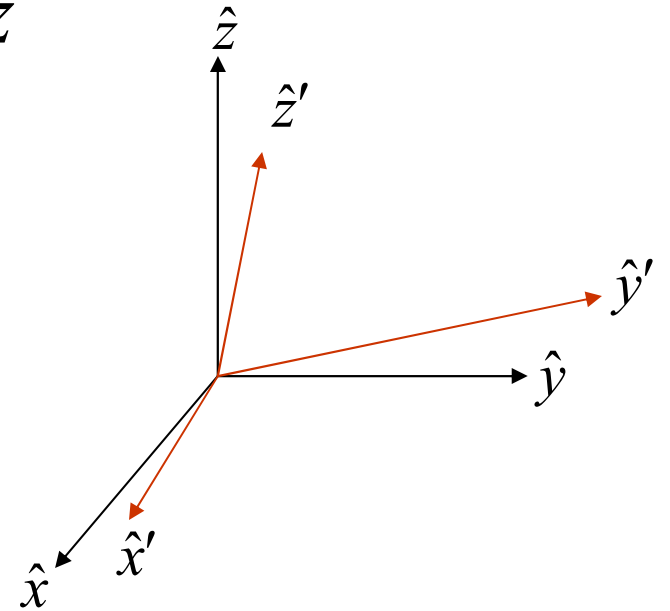
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A distortion of a material is described by the strain matrix

$$x' = (1 + \varepsilon_{xx})\hat{x} + \varepsilon_{xy}\hat{y} + \varepsilon_{xz}\hat{z}$$

$$y' = \varepsilon_{yx}\hat{x} + (1 + \varepsilon_{yy})\hat{y} + \varepsilon_{yz}\hat{z}$$

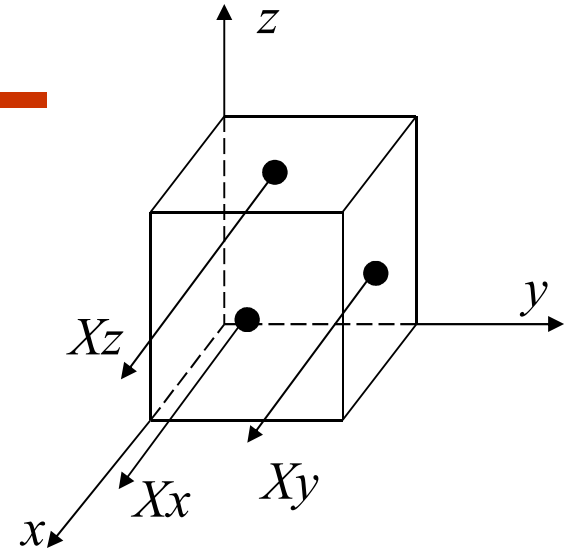
$$z' = \varepsilon_{zx}\hat{x} + \varepsilon_{zy}\hat{y} + (1 + \varepsilon_{zz})\hat{z}$$



# Stress

9 forces describe the stress

$X_x, X_y, X_z, Y_x, Y_y, Y_z, Z_x, Z_y, Z_z$



$X_x$  is a force applied in the  $x$ -direction to the plane normal to  $x$

$X_y$  is a shear force applied in the  $x$ -direction to the plane normal to  $y$

stress tensor:

$$\sigma = \begin{bmatrix} \frac{X_x}{A_x} & \frac{X_y}{A_y} & \frac{X_z}{A_z} \\ \frac{Y_x}{A_x} & \frac{Y_y}{A_y} & \frac{Y_z}{A_z} \\ \frac{Z_x}{A_x} & \frac{Z_y}{A_y} & \frac{Z_z}{A_z} \end{bmatrix}$$

Stress is force/m<sup>2</sup>

# Stress and Strain

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$$\boldsymbol{\varepsilon}_{ij} = S_{ijkl} \boldsymbol{\sigma}_{kl}$$

The stress - strain relationship is described by a rank 4 stiffness tensor. The inverse of the stiffness tensor is the compliance tensor.

$$\boldsymbol{\sigma}_{ij} = C_{ijkl} \boldsymbol{\varepsilon}_{kl}$$

Einstein convention: sum over repeated indices.

$$\begin{aligned} \varepsilon_{xx} = & S_{xxxx} \sigma_{xx} + S_{xxxy} \sigma_{xy} + S_{xxxz} \sigma_{xz} + S_{xxyx} \sigma_{yx} + S_{xxyy} \sigma_{yy} \\ & + S_{xxyz} \sigma_{yz} + S_{xxzx} \sigma_{zx} + S_{xxzy} \sigma_{zy} + S_{xxzz} \sigma_{zz} \end{aligned}$$

$$\begin{aligned}
d\epsilon_{ij} &= \left(\frac{\partial \epsilon_{ij}}{\partial \sigma_{kl}}\right) d\sigma_{kl} + \left(\frac{\partial \epsilon_{ij}}{\partial E_k}\right) dE_k + \left(\frac{\partial \epsilon_{ij}}{\partial H_l}\right) dH_l + \left(\frac{\partial \epsilon_{ij}}{\partial T}\right) dT \\
dP_i &= \left(\frac{\partial P_i}{\partial \sigma_{kl}}\right) d\sigma_{kl} + \left(\frac{\partial P_i}{\partial E_k}\right) dE_k + \left(\frac{\partial P_i}{\partial H_l}\right) dH_l + \left(\frac{\partial P_i}{\partial T}\right) dT \\
dM_i &= \left(\frac{\partial M_i}{\partial \sigma_{kl}}\right) d\sigma_{kl} + \left(\frac{\partial M_i}{\partial E_k}\right) dE_k + \left(\frac{\partial M_i}{\partial H_l}\right) dH_l + \left(\frac{\partial M_i}{\partial T}\right) dT \\
dS &= \left(\frac{\partial S}{\partial \sigma_{kl}}\right) d\sigma_{kl} + \left(\frac{\partial S}{\partial E_k}\right) dE_k + \left(\frac{\partial S}{\partial H_l}\right) dH_l + \left(\frac{\partial S}{\partial T}\right) dT
\end{aligned}$$

1. Elastic deformation.
2. Reciprocal (or converse) piezo-electric effect.
3. Reciprocal (or converse) piezo-magnetic effect.
4. Thermal dilatation.
5. Piezo-electric effect.
6. Electric polarization.
7. Magneto-electric polarization.
8. Pyroelectricity.
9. Piezo-magnetic effect.
10. Reciprocal (or converse) magneto-electric polarization.
11. Magnetic polarization.
12. Pyromagnetism.
13. Piezo-caloric effect.
14. Electro-caloric effect.
15. Magneto-caloric effect.
16. Heat transmission.

# Direct and reciprocal effects (Maxwell relations)

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$$-\left(\frac{\partial^2 G}{\partial \sigma_{ij} \partial E_k}\right) = \left(\frac{\partial P_k}{\partial \sigma_{ij}}\right) = -\left(\frac{\partial^2 G}{\partial E_k \partial \sigma_{ij}}\right) = \left(\frac{\partial \epsilon_{ij}}{\partial E_k}\right) = d_{kij}$$

$$-\left(\frac{\partial^2 G}{\partial \sigma_{ij} \partial H_l}\right) = \left(\frac{\partial M_l}{\partial \sigma_{ij}}\right) = -\left(\frac{\partial^2 G}{\partial H_l \partial \sigma_{ij}}\right) = \left(\frac{\partial \epsilon_{ij}}{\partial H_l}\right) = g_{lij}$$

$$-\left(\frac{\partial^2 G}{\partial E_k \partial H_l}\right) = \left(\frac{\partial M_l}{\partial E_k}\right) = -\left(\frac{\partial^2 G}{\partial H_l \partial E_k}\right) = \left(\frac{\partial P_k}{\partial H_l}\right) = \lambda_{lk}$$

$$-\left(\frac{\partial^2 G}{\partial \sigma_{ij} \partial T}\right) = \left(\frac{\partial S}{\partial \sigma_{ij}}\right) = -\left(\frac{\partial^2 G}{\partial T \partial \sigma_{ij}}\right) = \left(\frac{\partial \epsilon_{ij}}{\partial T}\right) = \alpha_{ij}$$

$$-\left(\frac{\partial^2 G}{\partial T \partial E_k}\right) = \left(\frac{\partial P_k}{\partial T}\right) = -\left(\frac{\partial^2 G}{\partial E_k \partial T}\right) = \left(\frac{\partial S}{\partial E_k}\right) = p_k$$

$$-\left(\frac{\partial^2 G}{\partial T \partial H_l}\right) = \left(\frac{\partial M_l}{\partial T}\right) = -\left(\frac{\partial^2 G}{\partial H_l \partial T}\right) = \left(\frac{\partial S}{\partial H_l}\right) = m_l.$$

Useful to check for errors in experiments or calculations

# Tensors

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Rank 2 tensors couple two vectors:

Electrical conductivity, thermal conductivity, electric susceptibility, magnetic susceptibility, Peltier effect (heat current due to electrical current), Seebeck effect (Electric field due to thermal gradient)

Rank 3 tensors couple a vector to a matrix:

Piezoelectricity, piezomagnetism

Rank 4 tensors couple two matrices:

Stiffness, compliance, piezoconductivity

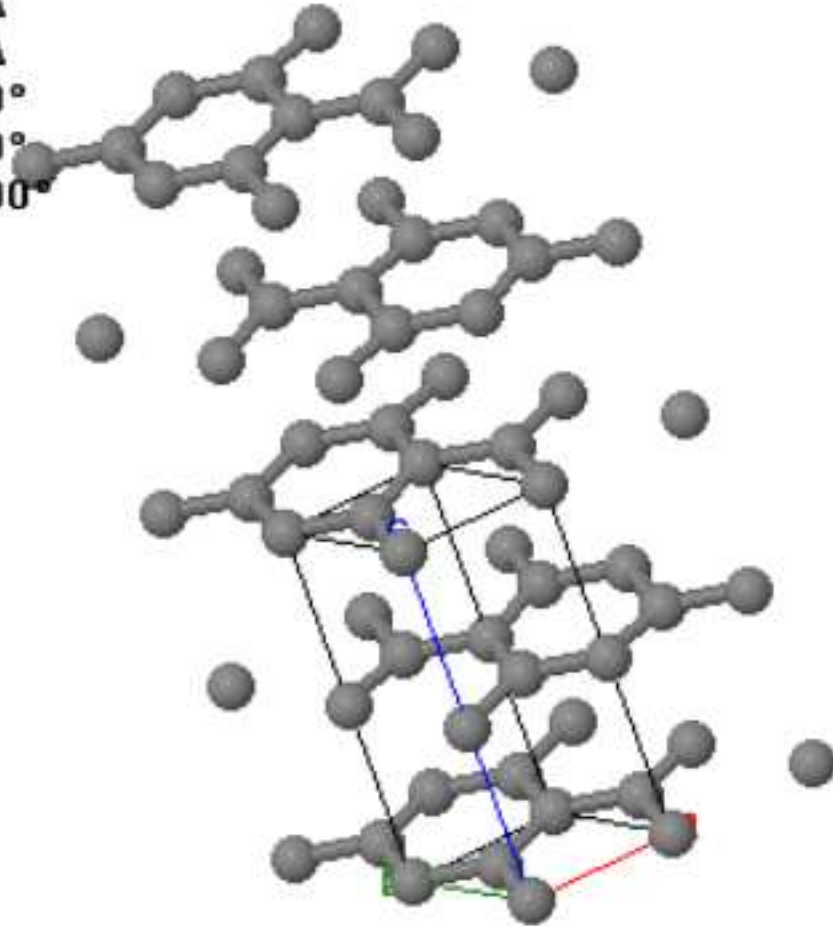
# graphite

Space group 194

4 inequivalent C atoms in the primitive unit cell

**Polytypes of carbon**  
graphite (hexagonal)  
graphene  
carbon nanotubes  
diamond  
rhombohedral graphite  
hexagonal diamond

HM: P 63 m c  
a=2.456Å  
b=2.456Å  
c=6.696Å  
α=90.000°  
β=90.000°  
γ=120.000°



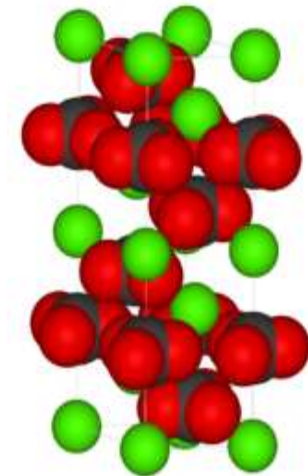


# Birefringence

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Calcite



name	international	Schoenflies	examples
rhombohedral holohedral	$\bar{3}m$	$D_{3d}$	calcite, corundum, hematite
rhombohedral hemimorphic	$3m$	$C_{3v}$	tourmaline, alunite
rhombohedral tetartohedral	$\bar{3}$	$S_6$	dolomite, ilmenite
trapezohedral	$32$	$D_3$	quartz, cinnabar
rhombohedral tetartohedral	$3$	$C_3$	none verified

# Nonlinear optics

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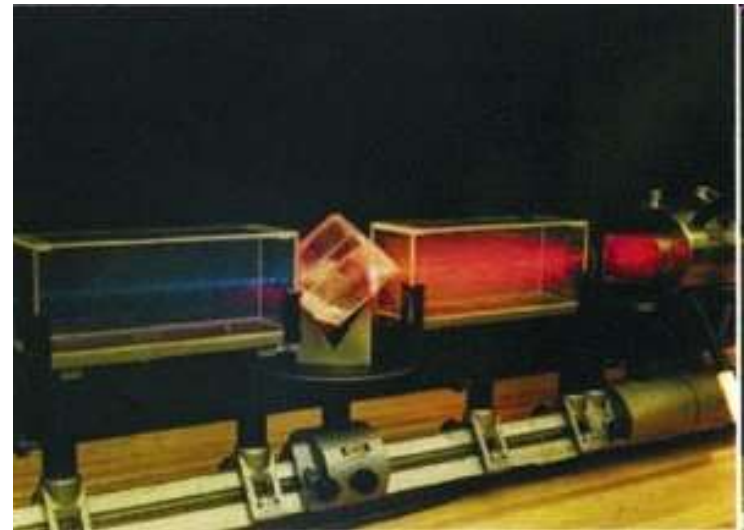
Period doubling crystals

no inversion symmetry

$$P = \chi^{(1)} E + \chi^{(2)} E^2 + \chi^{(3)} E^3 + \dots$$

$$P_i = \frac{-\partial^2 G}{\partial E_i \partial E_j} E_j + \frac{1}{2} \frac{-\partial^3 G}{\partial E_i \partial E_j \partial E_k} E_j E_k + \dots$$

$$\cos^2(\omega t) = \frac{1}{2}(1 + \cos(2\omega t))$$



806 nm light : lithium iodate ( $\text{LiIO}_3$ )

860 nm light : potassium niobate ( $\text{KNbO}_3$ )

980 nm light :  $\text{KNbO}_3$

1064 nm light : monopotassium phosphate ( $\text{KH}_2\text{PO}_4$ , KDP), lithium triborate (LBO).

1300 nm light : gallium selenide (GaSe)

1319 nm light :  $\text{KNbO}_3$ , BBO, KDP, lithium niobate ( $\text{LiNbO}_3$ ),  $\text{LiIO}_3$

# Electrostriction

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$$\frac{\partial P_k}{\partial \sigma_{ij}} = \frac{\partial \epsilon_{ij}}{\partial E_k} = - \left( \frac{\partial^2 G}{\partial E_k \partial \sigma_{ij}} \right) = d_{ijk}$$

$$\epsilon_{ij} = d_{ijk} E_k + Q_{ijkl} E_k E_l + \dots$$

piezoelectricity

Electrostriction

# Point Groups

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Crystals can have symmetries: rotation, reflection, inversion,...

$$\begin{pmatrix} x' \\ y' \\ z' \end{pmatrix} = \begin{pmatrix} 1 & 0 & 0 \\ 0 & \cos \alpha & \sin \alpha \\ 0 & -\sin \alpha & \cos \alpha \end{pmatrix} \begin{pmatrix} x \\ y \\ z \end{pmatrix}$$

Symmetries can be represented by matrices.

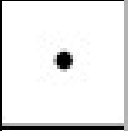


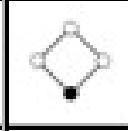
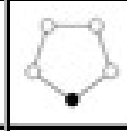
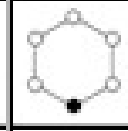
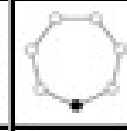
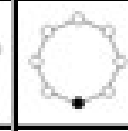
All such matrices that bring the crystal into itself form the group of the crystal.

$$AB \in G \text{ for } A, B \in G$$

32 point groups (one point remains fixed during transformation)

230 space groups

# Cyclic groups

							
$C_1$	$C_2$	$C_3$	$C_4$	$C_5$	$C_6$	$C_7$	$C_8$

$$C_2 \quad E = \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_2 = \begin{bmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & 1 \end{bmatrix}$$

$$C_4 \quad E = \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_4 = \begin{bmatrix} 0 & -1 & 0 \\ 1 & 0 & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_2 = \begin{bmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_4^3 = \begin{bmatrix} 0 & 1 & 0 \\ -1 & 0 & 0 \\ 0 & 0 & 1 \end{bmatrix}$$

$$E = \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_6 = \begin{bmatrix} \frac{1}{2} & -\frac{\sqrt{3}}{2} & 0 \\ \frac{\sqrt{3}}{2} & \frac{1}{2} & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_3 = \begin{bmatrix} -\frac{1}{2} & -\frac{\sqrt{3}}{2} & 0 \\ \frac{\sqrt{3}}{2} & -\frac{1}{2} & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_2 = \begin{bmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_3^2 = \begin{bmatrix} -\frac{1}{2} & \frac{\sqrt{3}}{2} & 0 \\ -\frac{\sqrt{3}}{2} & -\frac{1}{2} & 0 \\ 0 & 0 & 1 \end{bmatrix}, C_6^5 = \begin{bmatrix} \frac{1}{2} & \frac{\sqrt{3}}{2} & 0 \\ -\frac{\sqrt{3}}{2} & \frac{1}{2} & 0 \\ 0 & 0 & 1 \end{bmatrix}$$

[http://en.wikipedia.org/wiki/Cyclic\\_group](http://en.wikipedia.org/wiki/Cyclic_group)

# Pyroelectricity

$$\pi_i = - \left( \frac{\partial^2 G}{\partial E_i \partial T} \right)$$

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Pyroelectricity is described by a rank 1 tensor

$$\pi_i = \frac{\partial P_i}{\partial T}$$

$$\begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & -1 \end{bmatrix} \begin{bmatrix} \pi_x \\ \pi_y \\ \pi_z \end{bmatrix} = \begin{bmatrix} \pi_x \\ \pi_y \\ -\pi_z \end{bmatrix} \Rightarrow \begin{bmatrix} \pi_x \\ \pi_y \\ 0 \end{bmatrix}$$

$$\begin{bmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & -1 \end{bmatrix} \begin{bmatrix} \pi_x \\ \pi_y \\ \pi_z \end{bmatrix} = \begin{bmatrix} -\pi_x \\ -\pi_y \\ -\pi_z \end{bmatrix} \Rightarrow \begin{bmatrix} 0 \\ 0 \\ 0 \end{bmatrix}$$

# Pyroelectricity

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Quartz, ZnO, LaTaO<sub>3</sub>

**example**

Turmalin: point group 3m  
for  $\Delta T = 1^\circ\text{C}$ ,  
 $\Delta E \sim 7 \cdot 10^4 \text{ V/m}$

Pyroelectrics have a spontaneous polarization. If it can be reversed by an electric field they are called Ferroelectrics (BaTiO<sub>3</sub>)

Pyroelectrics are at Joanneum research to make infrared detectors (to detect humans).

10 Pyroelectric crystal classes: 1, 2, m, mm2, 3, 3m, 4, 4mm, 6, 6mm

# Electric susceptibility $\chi_{ij} = -\left(\frac{\partial^2 G}{\partial E_i \partial E_j}\right)$

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$$P_i = \chi_{ij} E_j$$

$$\begin{bmatrix} P_x \\ P_y \\ P_z \end{bmatrix} = \begin{bmatrix} \chi_{xx} & \chi_{xy} & \chi_{xz} \\ \chi_{yx} & \chi_{yy} & \chi_{yz} \\ \chi_{zx} & \chi_{zy} & \chi_{zz} \end{bmatrix} \begin{bmatrix} E_x \\ E_y \\ E_z \end{bmatrix}$$

Transforming  $P$  and  $E$  by a crystal symmetry must leave the susceptibility tensor unchanged

$$U\vec{P} = \chi U\vec{E}$$

$$U^{-1}U\vec{P} = U^{-1}\chi U\vec{E}$$

$$\chi = U^{-1}\chi U$$

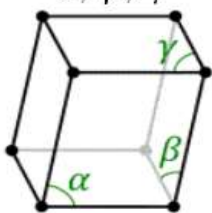
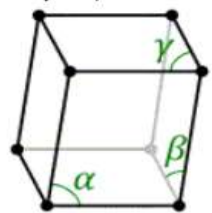
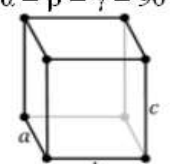
If rotation by 180 about the  $z$  axis is a symmetry,

$$U = \begin{bmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & 1 \end{bmatrix} \quad U^{-1} = U = \begin{bmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & 1 \end{bmatrix} \quad U^{-1}\chi U = \begin{bmatrix} \chi_{xx} & \chi_{xy} & -\chi_{xz} \\ \chi_{yx} & \chi_{yy} & -\chi_{yz} \\ -\chi_{zx} & -\chi_{zy} & \chi_{zz} \end{bmatrix}$$

$$\chi_{xz} = \chi_{yz} = \chi_{zx} = \chi_{zy} = 0$$



## The 32 Crystal Classes

Crystal system	Crystal Class	International symbol	Schoenflies symbol	Space groups	2-fold axes	3-fold axes	4-fold axes	6-fold axes	mirror planes	inversion	Examples	N sym ele
<b>Triclinic</b> $a \neq b \neq c$ $\alpha \neq \beta \neq \gamma$ 	triclinic-pedial	1	$C_1$	1	-	-	-	-	-	n		
	triclinic-pinacoidal	$\bar{1}$	$S_2 = C_i$	2	-	-	-	-	-	y		
<b>Monoclinic</b> $a \neq b \neq c$ $\alpha \neq 90^\circ$ , $\beta = \gamma = 90^\circ$ 	monoclinic-sphenoidal	2	$C_2$	3-5	1	-	-	-	-	n		
	monoclinic-domatic	$m$	$C_{1h} = C_s$	6-9	-	-	-	-	1	n		
	monoclinic-prismatic	$2/m$	$C_{2h}$	10-15	1	-	-	-	1	y		
<b>Orthorhombic</b> $a \neq b \neq c$ $\alpha = \beta = \gamma = 90^\circ$ 	orthorhombic-disphenoidal	222	$V = D_2$	16-24	3	-	-	-	-	n		
	orthorhombic-pyramidal	$mm2$	$C_{2v}$	25-46	1	-	-	-	2	n		
											47: $YBa_2Cu_3O_{7-x}$	

# Symmetric Tensors

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$$\chi_{ij}^E = \frac{\partial P_i}{\partial E_j} = -\frac{\partial^2 G}{\partial E_i \partial E_j} = \frac{\partial P_j}{\partial E_i} = \chi_{ji}^E$$

$$\begin{bmatrix} g_{11} & g_{12} & g_{13} \\ g_{12} & g_{22} & g_{23} \\ g_{13} & g_{23} & g_{33} \end{bmatrix}$$

# Tensor notation

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We need a way to represent 3rd and 4th rank tensors in 2-d.

$$1\ 1 \rightarrow 1 \quad 1\ 2 \rightarrow 6 \quad 1\ 3 \rightarrow 5$$

$$2\ 2 \rightarrow 2 \quad 2\ 3 \rightarrow 4$$

$$3\ 3 \rightarrow 3$$

rank 3

$$\mathcal{g}_{36} \rightarrow \mathcal{g}_{312}$$

rank 4

$$\mathcal{g}_{14} \rightarrow \mathcal{g}_{1123}$$

# Cubic crystals

All second rank tensors of cubic crystals reduce to constants

Electrical conductivity, thermal conductivity, electric susceptibility, magnetic susceptibility, Peltier effect (heat current due to electrical current), Seebeck effect (Electric field due to thermal gradient)

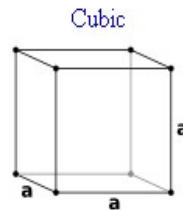
216: ZnS, GaAs, GaP, InAs, SiC

221: CsCl, cubic perovskite

225: Al, Cu, Ni, Ag, Pt, Au, Pb, NaCl

227: C, Si, Ge, spinel

229: Na, K, Cr, Fe, Nb, Mo, Ta



23	$T$	195-199		12
$m\bar{3}$	$T_h$	200-206		24
432	$O$	207-214		24
$\bar{4}3m$	$T_d$	215-220	216: Zincblende, ZnS, GaAs, GaP, InAs, SiC	24
$m\bar{3}m$	$O_h$	221-230	221: CsCl, cubic perovskite 225: fcc, Al, Cu, Ni, Ag, Pt, Au, Pb, $\gamma$ -Fe, NaCl 227: diamond, C, Si,	48

$$\begin{bmatrix} \xi_{11} & 0 & 0 \\ & \xi_{11} & 0 \\ & & \xi_{11} \end{bmatrix}$$

Material	↕	$\rho$ ( $\Omega \cdot m$ ) at 20 °C	$\sigma$ (S/m) at 20 °C	Temperature coefficient <sup>[note 1]</sup> ( $K^{-1}$ )	Reference
Silver		$1.59 \times 10^{-8}$	$6.30 \times 10^7$	0.0038	[7][8]
Copper		$1.68 \times 10^{-8}$	$5.96 \times 10^7$	0.0039	[8]
Annealed copper <sup>[note 2]</sup>		$1.72 \times 10^{-8}$	$5.80 \times 10^7$		[citation needed]
Gold <sup>[note 3]</sup>		$2.44 \times 10^{-8}$	$4.10 \times 10^7$	0.0034	[7]
Aluminium <sup>[note 4]</sup>		$2.82 \times 10^{-8}$	$3.5 \times 10^7$	0.0039	[7]
Calcium		$3.36 \times 10^{-8}$	$2.98 \times 10^7$	0.0041	
Tungsten		$5.60 \times 10^{-8}$	$1.79 \times 10^7$	0.0045	[7]
Zinc		$5.90 \times 10^{-8}$	$1.69 \times 10^7$	0.0037	[9]
Nickel		$6.99 \times 10^{-8}$	$1.43 \times 10^7$	0.006	
Lithium		$9.28 \times 10^{-8}$	$1.08 \times 10^7$	0.006	
Iron		$1.0 \times 10^{-7}$	$1.00 \times 10^7$	0.005	[7]
Platinum		$1.06 \times 10^{-7}$	$9.43 \times 10^6$	0.00392	[7]
Tin		$1.09 \times 10^{-7}$	$9.17 \times 10^6$	0.0045	
Carbon steel (1010)		$1.43 \times 10^{-7}$	$6.99 \times 10^6$		[10]
Lead		$2.2 \times 10^{-7}$	$4.55 \times 10^6$	0.0039	[7]
Titanium		$4.20 \times 10^{-7}$	$2.38 \times 10^6$	X	
Grain oriented electrical steel		$4.60 \times 10^{-7}$	$2.17 \times 10^6$		[11]
Manganin		$4.82 \times 10^{-7}$	$2.07 \times 10^6$	0.000002	[12]
Constantan		$4.9 \times 10^{-7}$	$2.04 \times 10^6$	0.000008	[13]
Stainless steel <sup>[note 5]</sup>		$6.9 \times 10^{-7}$	$1.45 \times 10^6$		[14]
Mercury		$9.8 \times 10^{-7}$	$1.02 \times 10^6$	0.0009	[12]
Nichrome <sup>[note 6]</sup>		$1.10 \times 10^{-6}$	$9.09 \times 10^5$	0.0004	[7]
GaAs		$5 \times 10^{-7}$ to $10 \times 10^{-3}$	$5 \times 10^{-8}$ to $10^3$		[15]
Carbon (amorphous)		$5 \times 10^{-4}$ to $8 \times 10^{-4}$	$1.25$ to $2 \times 10^3$	-0.0005	[7][16]
Carbon (graphite) <sup>[note 7]</sup>		$2.5 \times 10^{-6}$ to $5.0 \times 10^{-6}$ //basal plane $3.0 \times 10^{-3}$ $\perp$ basal plane	$2$ to $3 \times 10^5$ //basal plane $3.3 \times 10^2$ $\perp$ basal plane		[17]
Carbon (diamond) <sup>[note 8]</sup>		$1 \times 10^{12}$	$\sim 10^{-13}$		[18]
Germanium <sup>[note 8]</sup>		$4.6 \times 10^{-1}$	2.17	-0.048	[7][8]
Sea water <sup>[note 9]</sup>		$2 \times 10^{-1}$	4.8		[19]
Sea water <sup>[note 10]</sup>		$2.4 \times 10^{-1}$ to $2.4 \times 10^{-3}$	$5.4 \times 10^{-4}$ to $5.4 \times 10^{-2}$		[citation needed]

# Piezoelectricity (rank 3 tensor)

AFM's, STM's

Quartz crystal oscillators

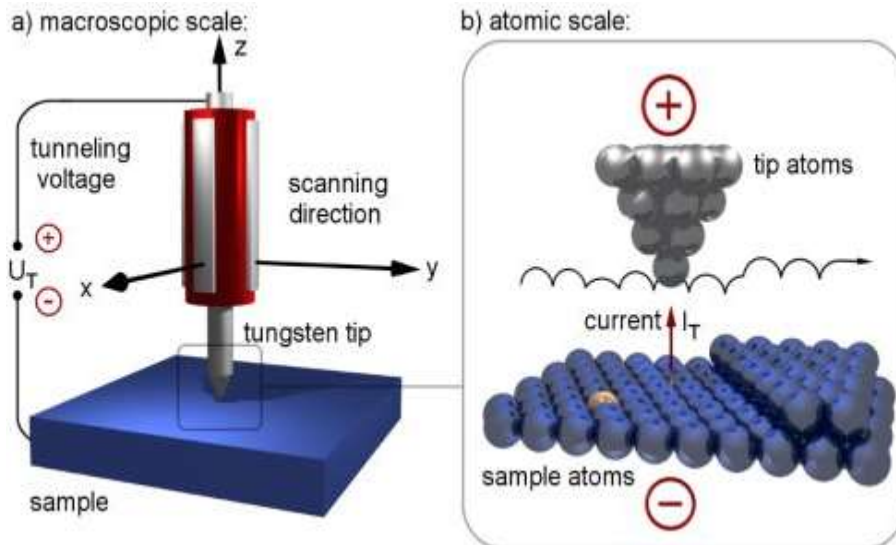
Surface acoustic wave generators

Pressure sensors - Epcos

Fuel injectors - Bosch

Inkjet printers

No inversion symmetry



lead zirconate titanate ( $\text{Pb}[\text{Zr}_x\text{Ti}_{1-x}]\text{O}_3$   $0 < x < 1$ )

—more commonly known as PZT

barium titanate ( $\text{BaTiO}_3$ )

lead titanate ( $\text{PbTiO}_3$ )

potassium niobate ( $\text{KNbO}_3$ )

lithium niobate ( $\text{LiNbO}_3$ )

lithium tantalate ( $\text{LiTaO}_3$ )

sodium tungstate ( $\text{Na}_2\text{WO}_3$ )

$\text{Ba}_2\text{NaNb}_5\text{O}_{15}$

$\text{Pb}_2\text{KNb}_5\text{O}_{15}$

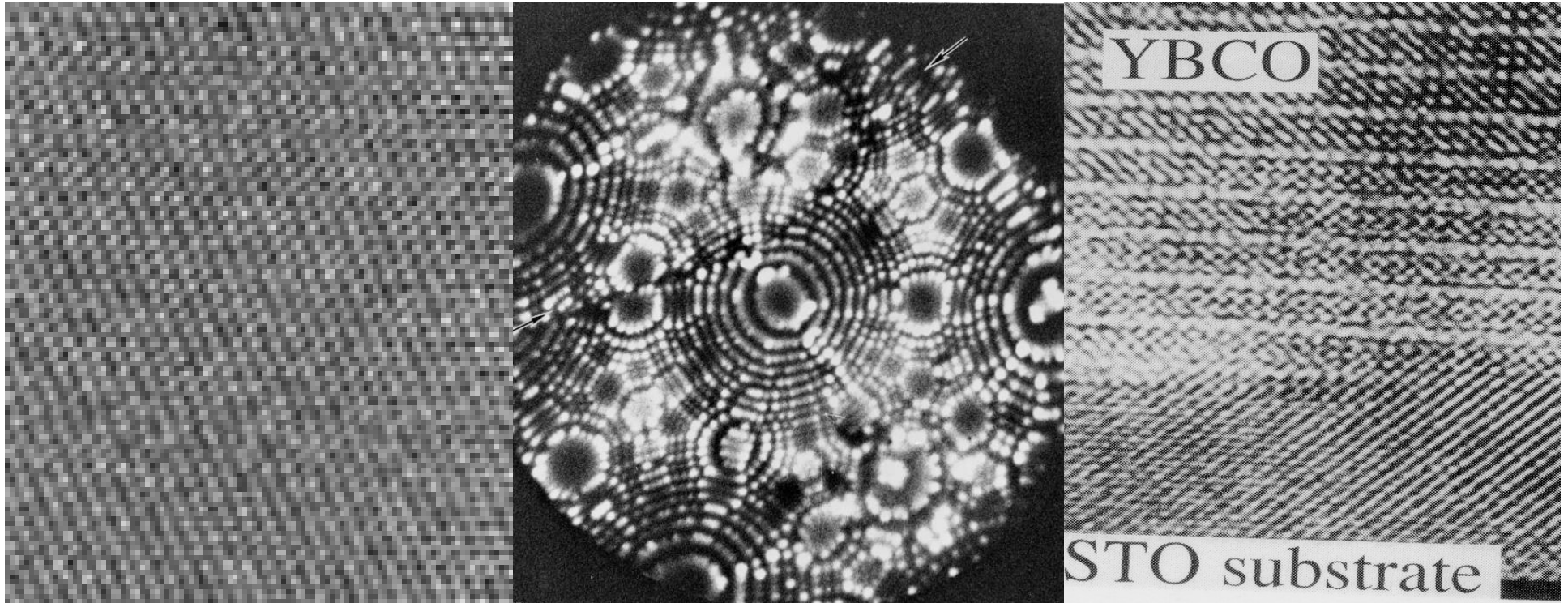
Piezoelectric crystal classes: 1, 2, m, 222, mm2, 4, -4, 422, 4mm, -42m, 3, 32, 3m, 6, -6, 622, 6mm, -62m, 23, -43m





# Crystal structure determination

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Scanning tunneling  
microscope

Field ion microscope

Transmission electron  
microscope

Usually x-ray diffraction is used to  
determine the crystal structure

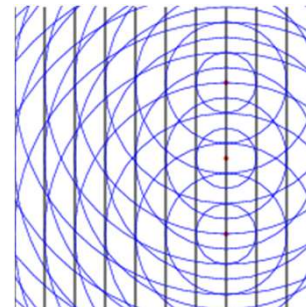
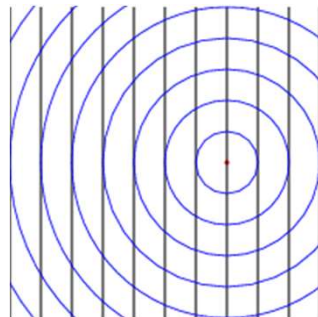


# Crystal diffraction (Beugung)

Everything moves like a wave but exchanges energy and momentum as a particle

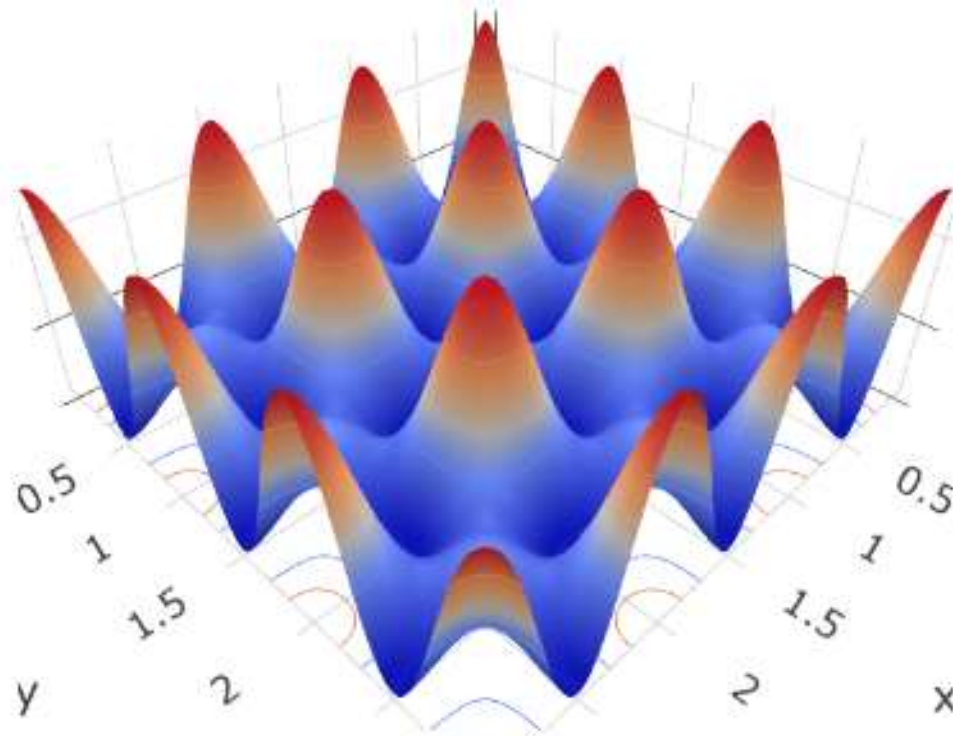
light  
sound  
electron waves  
neutron waves  
positron waves  
plasma waves

photons  
phonons  
electrons  
neutrons  
positrons  
plasmons



# Periodic functions

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Use a Fourier series to describe periodic functions