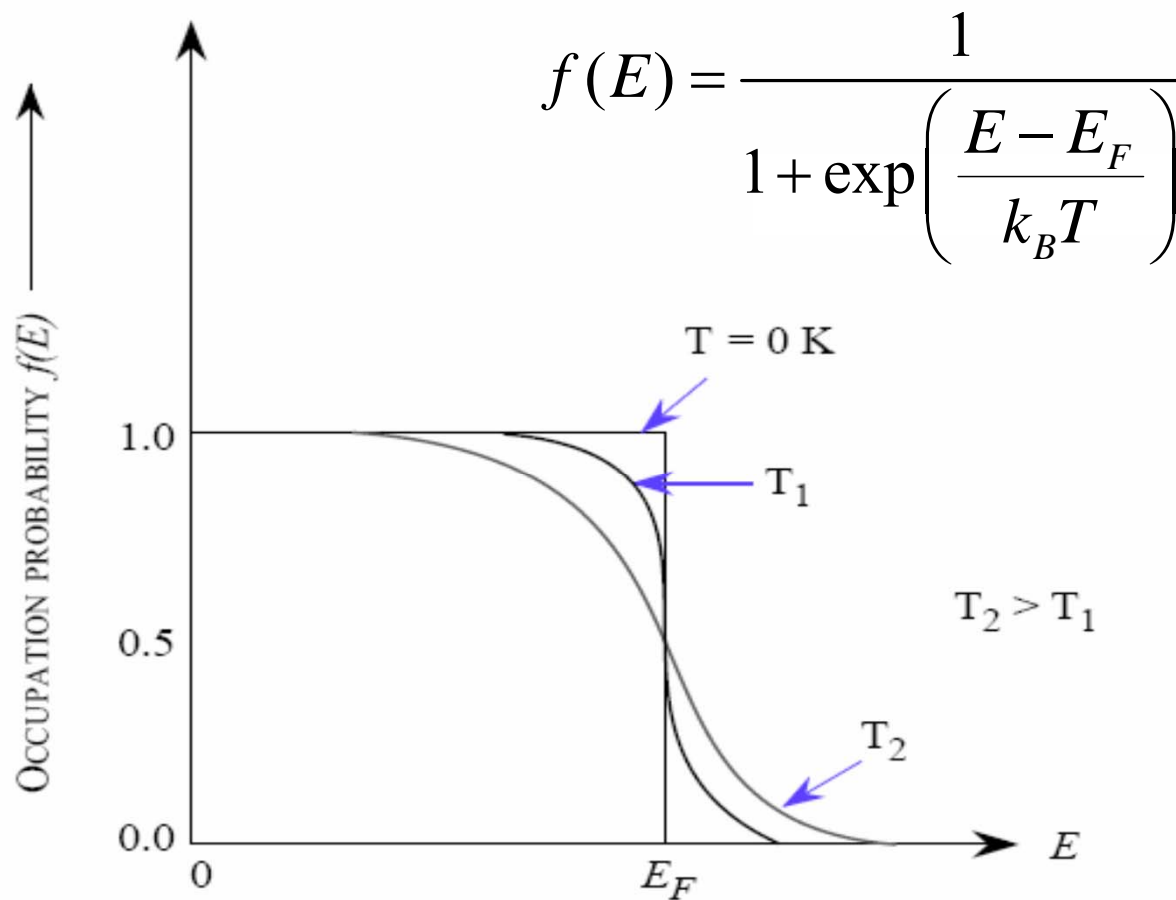


Intrinsic Semiconductors

Fermi function

$f(E)$ is the probability that a state at energy E is occupied.



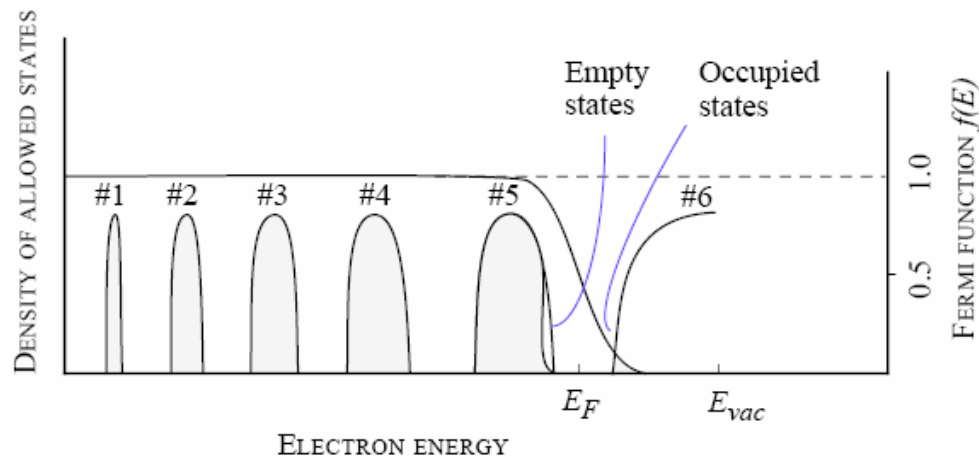
Fermi energy

The Fermi energy is implicitly defined as the energy that solves the following equation.

$$n = \int_{-\infty}^{\infty} D(E) f(E) dE$$

Here n is the electron density.

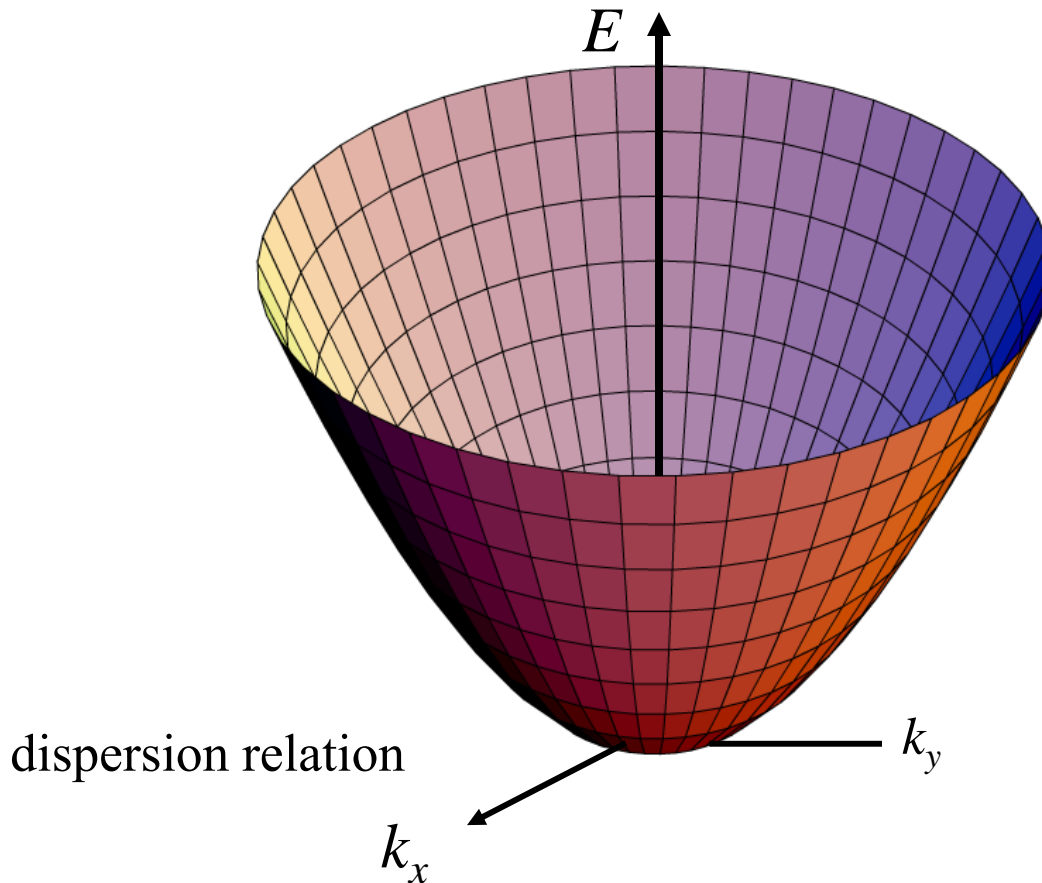
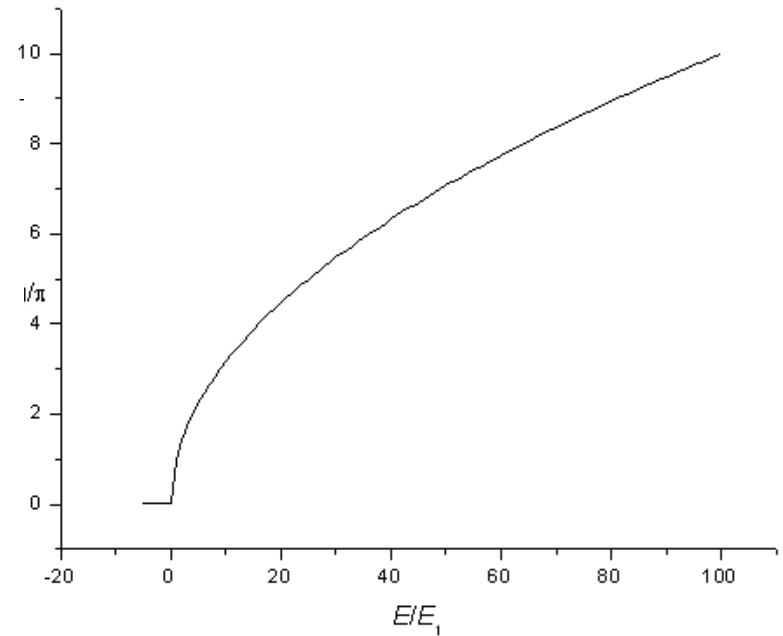
The density of states, the total number of electrons and the temperature are given. To find the Fermi energy, guess one and evaluate the integral. If n turns out too low, guess a higher E_F and if n turns out too high, guess a lower E_F .



free electrons (simple model for a metal)

$$E(\vec{k}) = \frac{1}{2}mv^2 = \frac{p^2}{2m} = \frac{\hbar^2}{2m}(k_x^2 + k_y^2 + k_z^2)$$

3-d density of states



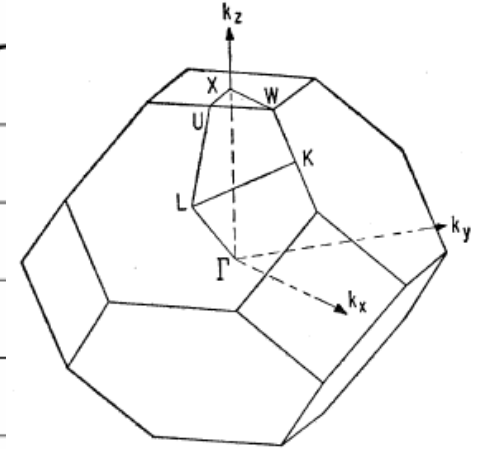
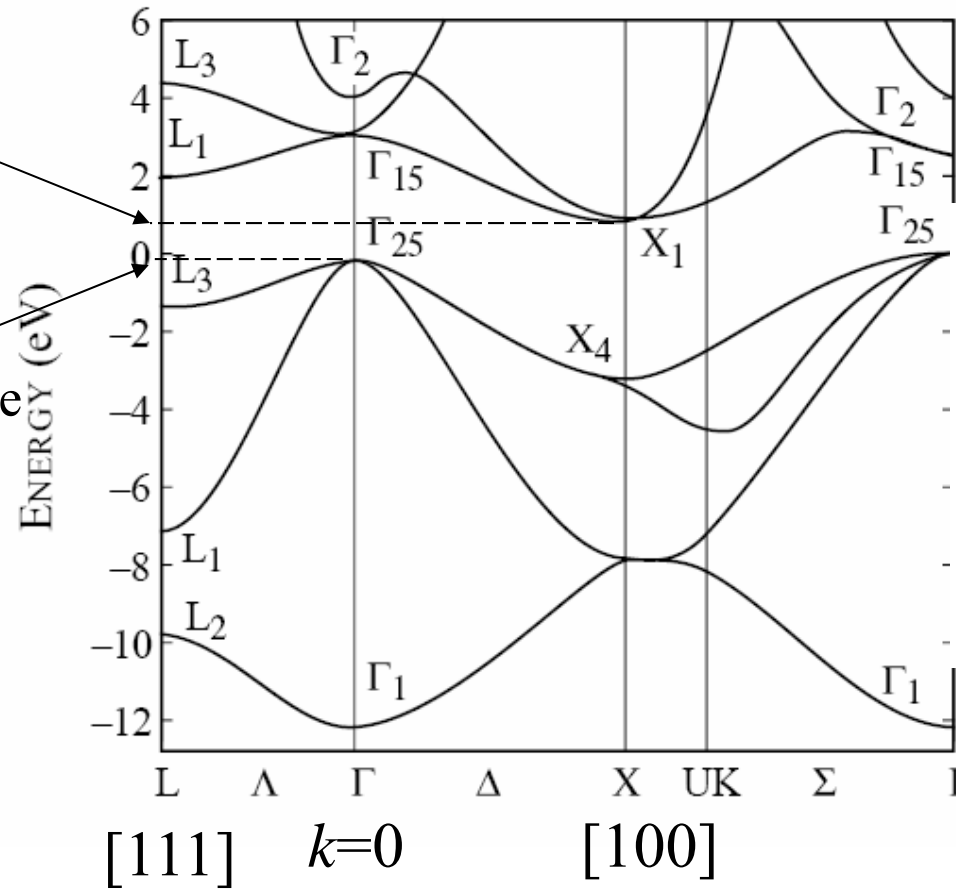
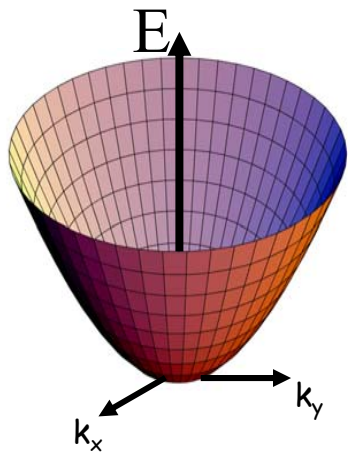
$$D(E) = \begin{cases} 0 & \text{for } E < 0 \\ \frac{(2m)^{3/2}}{2\pi^2\hbar^3} \sqrt{E} & \text{for } E > 0 \end{cases}$$

Silicon band structure

E_c = bottom of the conduction band

$$E_g = E_c - E_v$$

E_v = top of the valence band



Near the bottom of the conduction band, the band structure looks like a parabola.

Effective mass

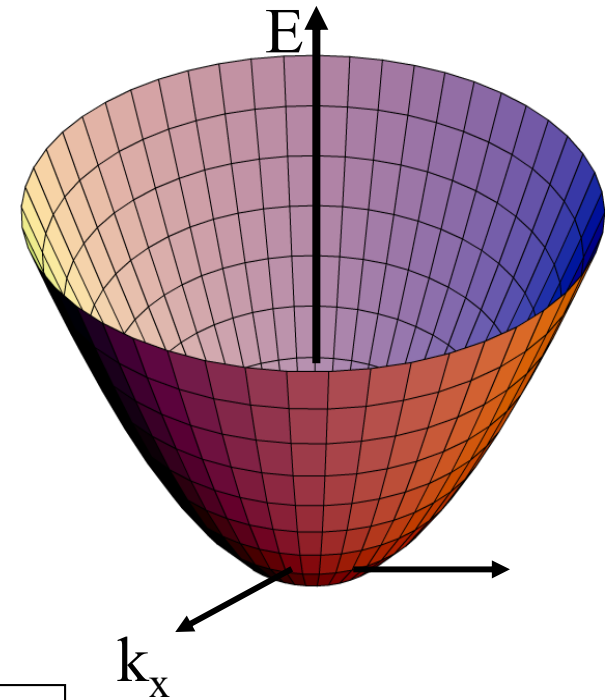
$$E(\vec{k}) = \frac{\hbar^2}{2m} (k_x^2 + k_y^2 + k_z^2) = \frac{p^2}{2m} = \frac{1}{2} m v^2$$

$$\frac{dE(\vec{k})}{dk_x} = \frac{\hbar^2 k_x}{m}$$

$$\frac{d^2 E(\vec{k})}{dk_x^2} = \frac{\hbar^2}{m}$$

Effective mass

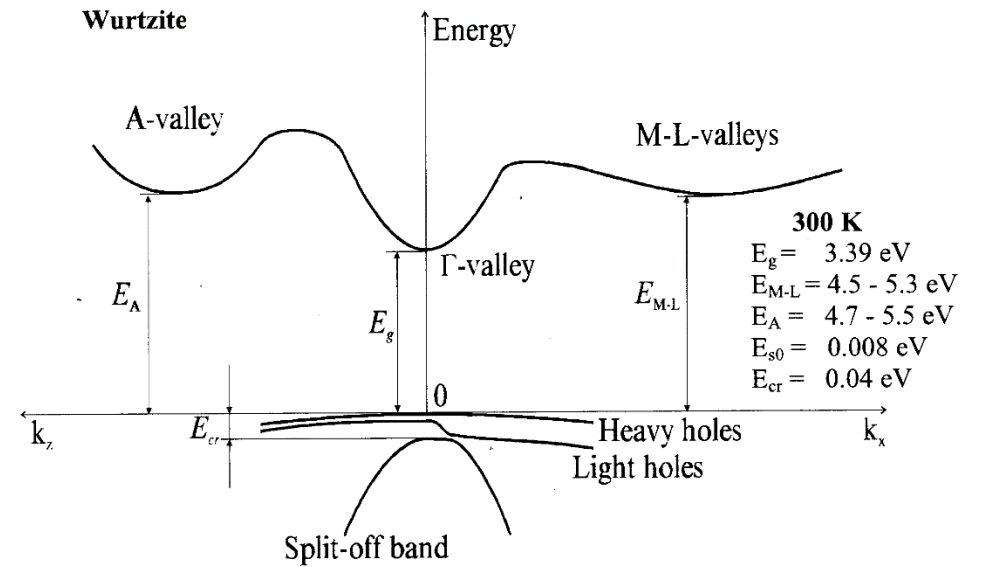
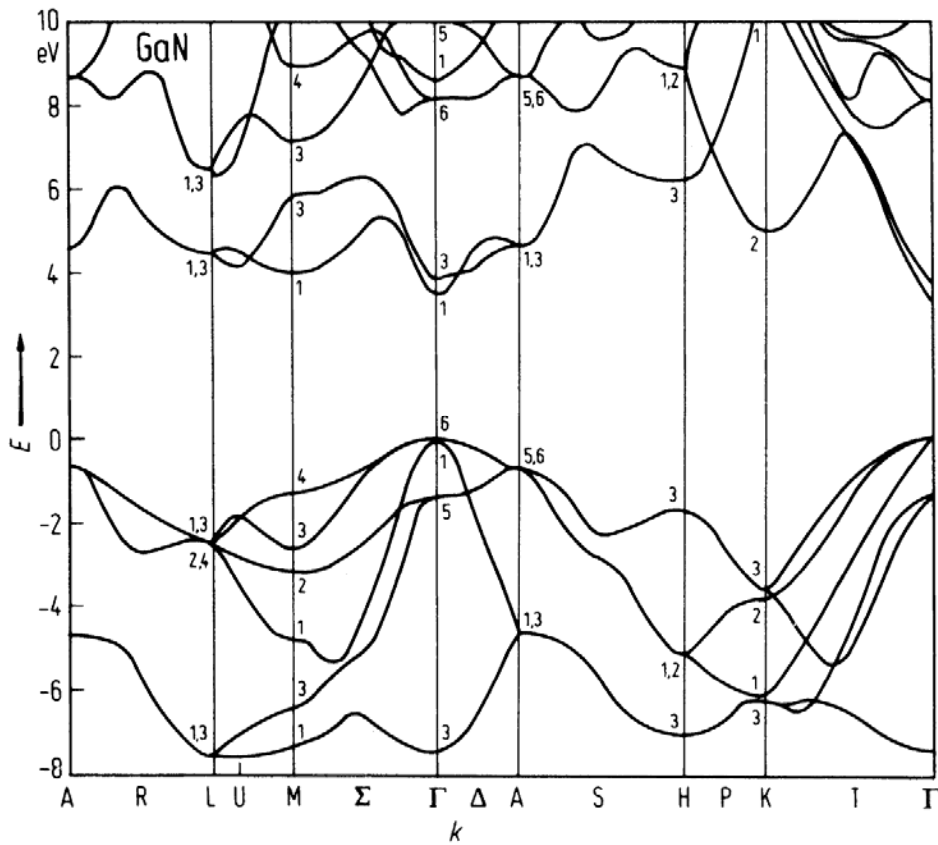
$$m_x^* = \frac{\hbar^2}{\frac{d^2 E(\vec{k})}{dk_x^2}}$$



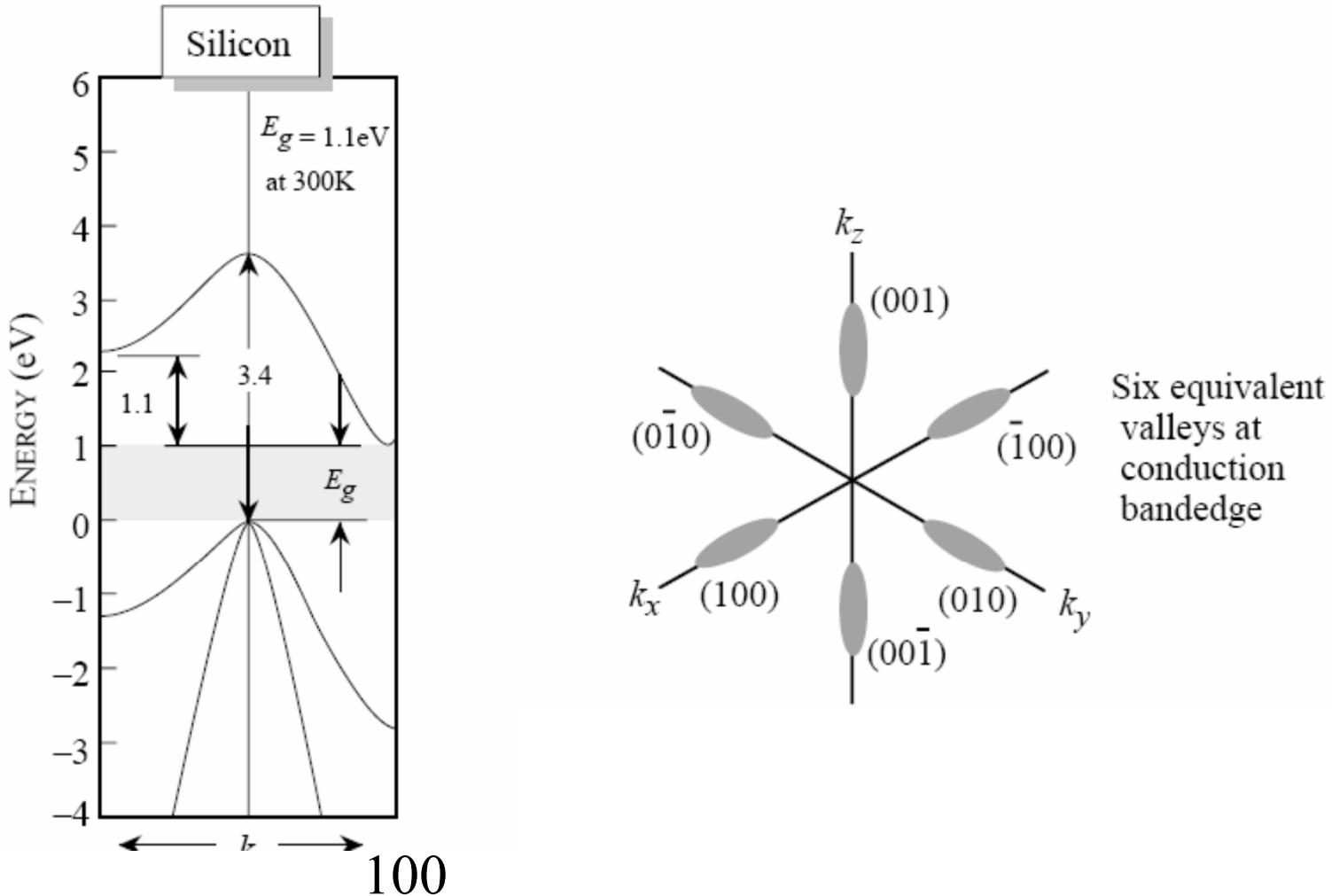
This effective mass is used to describe the response of electrons to external forces in the particle picture.

$$\vec{F} = -e\vec{E} = m^* \vec{a}$$

GaN



Anisotropic effective mass in silicon



The electrons seem to have different masses when the electric field is applied in different directions.

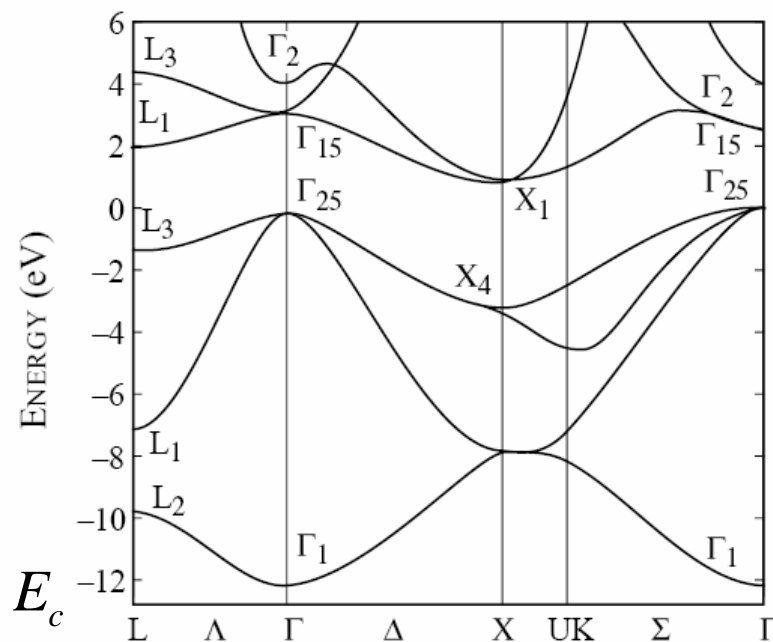
Return to
problem
list

Conduction band electron energy

In silicon, the bottom of the conduction valley along the (100) direction is at $(2\pi/a)(0.85, 0, 0)$ where $a = 0.543$ nm. Electrons in this valley have an anisotropic effective mass. The effective mass in the (100) direction is $m_l^* = 0.98m_0$ and the effective mass transverse to the [100] direction is $m_t^* = 0.19m_0$. What is the energy of an electron with a k -vector $(2\pi/a)(0.92, -0.01, 0.15)$?

$$E = \text{[input box]} \text{ eV}$$

$$E = \frac{\hbar^2 \left(k_x - 0.85 \frac{2\pi}{a} \right)^2}{2m_l} + \frac{\hbar^2 k_y^2}{2m_t} + \frac{\hbar^2 k_z^2}{2m_t} + E_c$$



Holes

When all states in a band are occupied, the band does not contribute to the current. There are as many left-moving electrons as right-moving electrons.

$$I \propto \sum_{\text{occupied } \vec{k}} (-e\vec{v}_{\vec{k}})$$

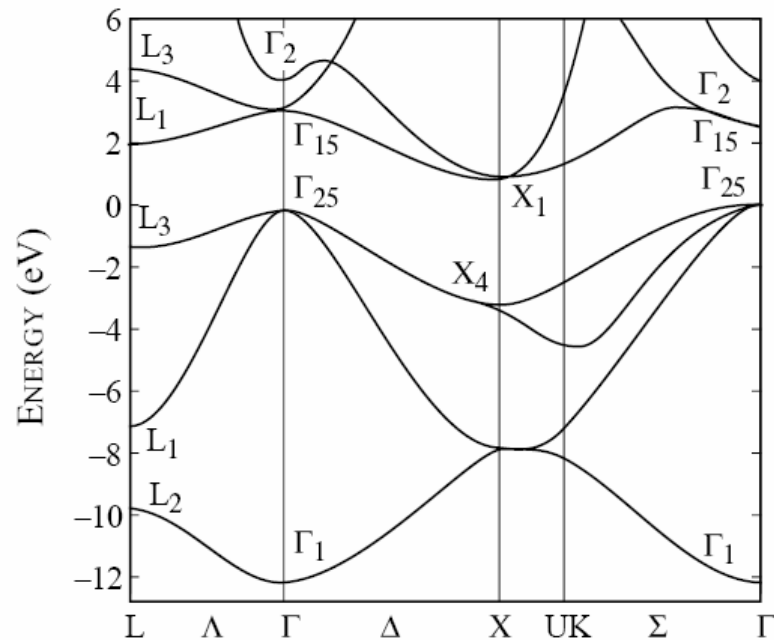
$$I \propto \sum_{\text{all } \vec{k}} (-e\vec{v}_{\vec{k}}) - \sum_{\text{empty } \vec{k}} (-e\vec{v}_{\vec{k}})$$

$$I \propto \sum_{\text{empty } \vec{k}} e\vec{v}_{\vec{k}}$$

valence band, holes

In the valence band, the effective mass is negative.

$$m^* = \frac{\hbar^2}{\frac{d^2 E(\vec{k})}{dk_x^2}} < 0$$



Holes

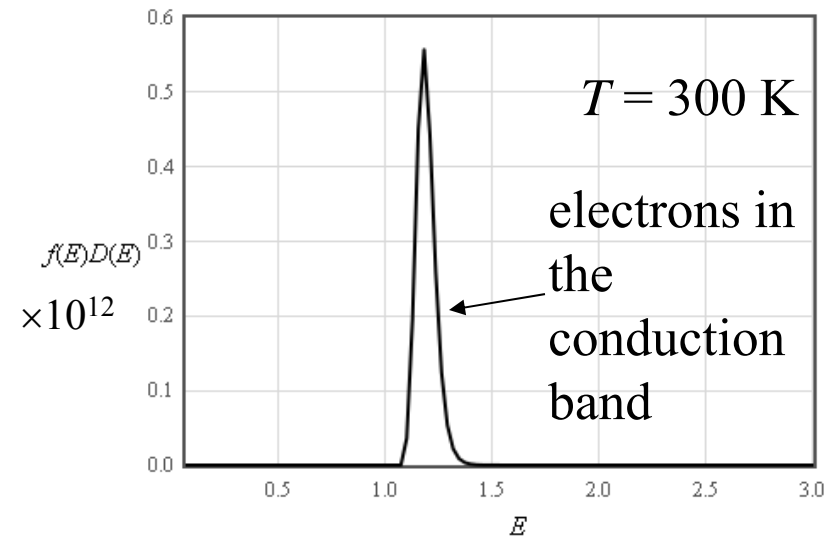
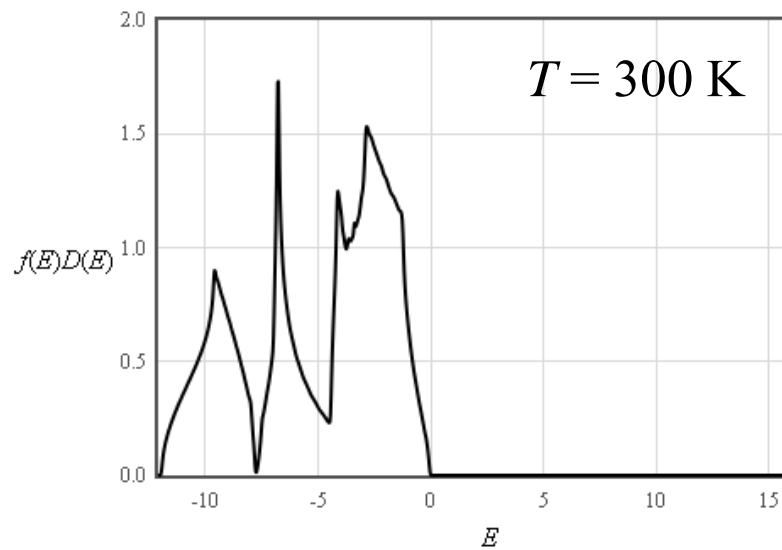
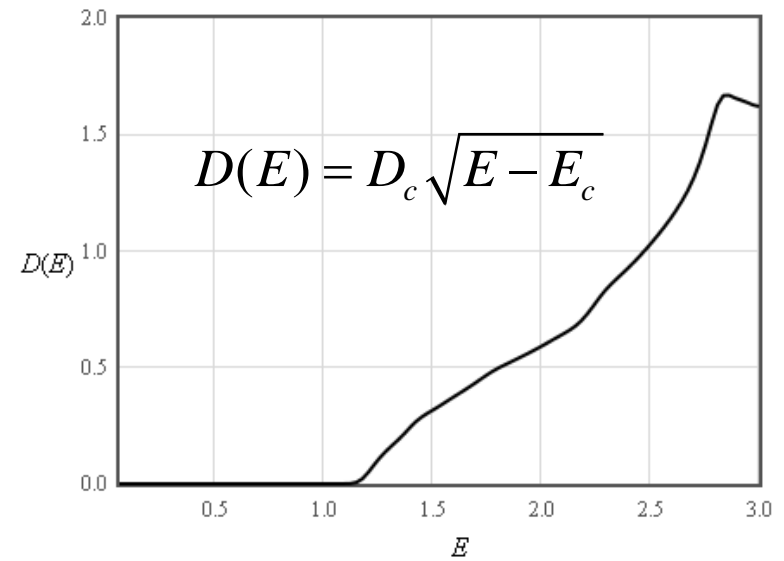
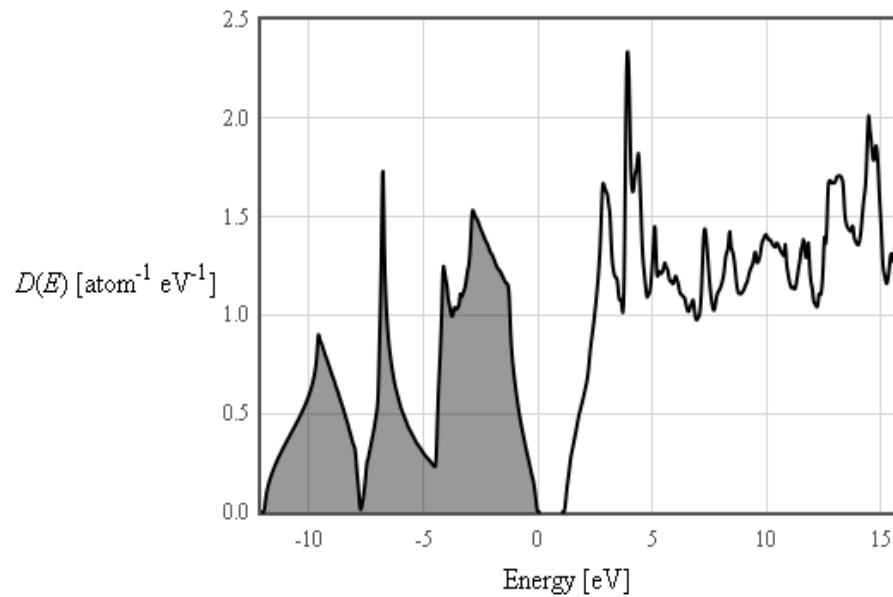
Charge carriers in the valence band can be considered to be positively charged holes. The number of holes in the valence band is the number of missing electrons.

m_h^* = effective mass of holes

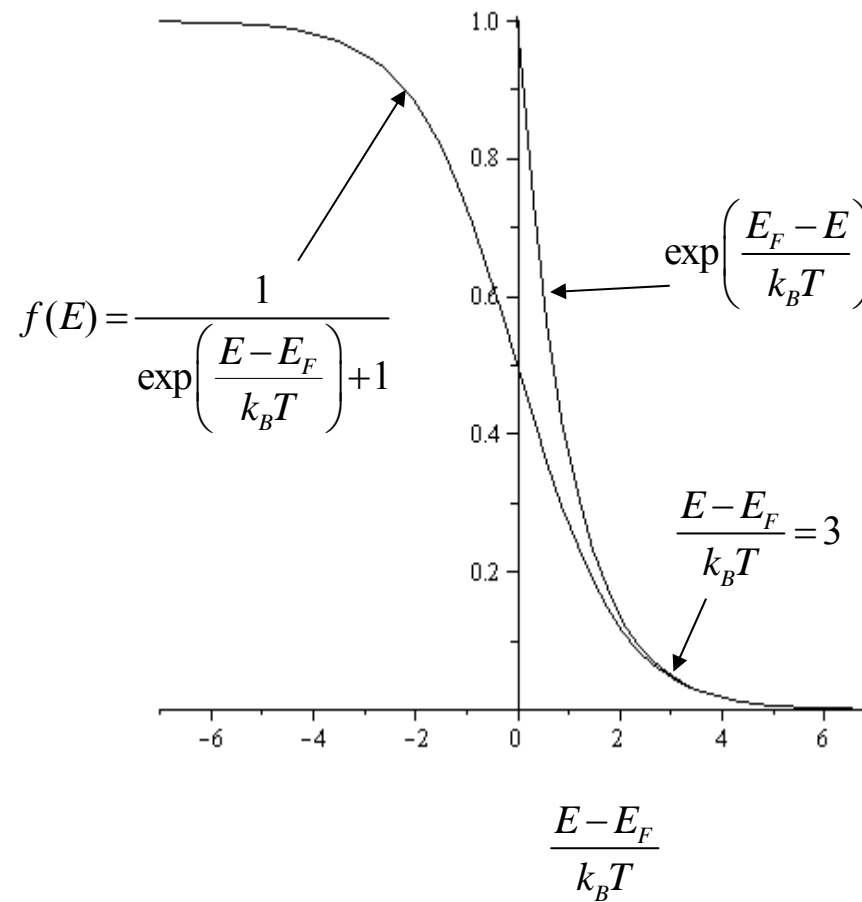
$$m_h^* = - \frac{\hbar^2}{\frac{d^2 E(\vec{k})}{dk_x^2}}$$

$$\vec{F} = e\vec{E} = m_h^* \vec{a}$$

Silicon density of states



Boltzmann approximation



Density of electrons in the conduction band

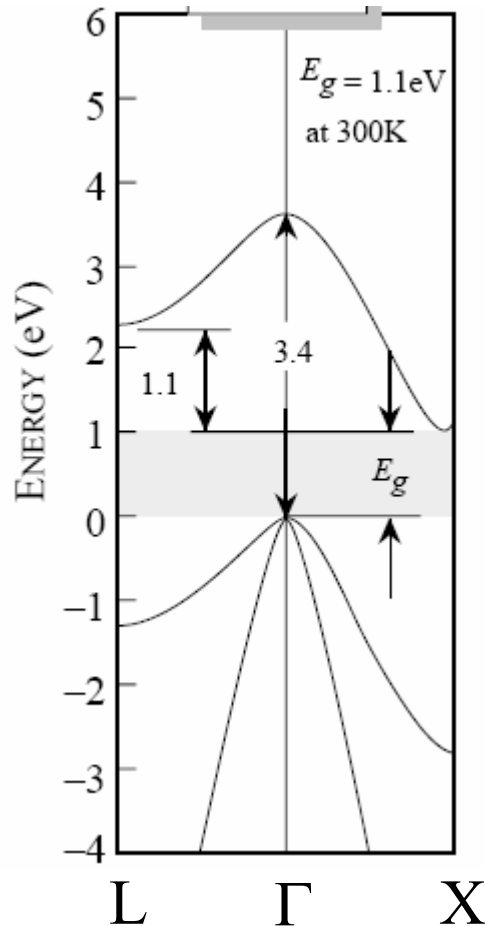
$$n = \int_{E_c}^{\infty} D(E) f(E) dE \approx D_c \int_{E_c}^{\infty} \exp\left(\frac{E_F - E}{k_B T}\right) \sqrt{E - E_c} dE$$

$$x = E - E_c \quad \int_0^{\infty} \sqrt{x} \exp\left(\frac{-x}{k_B T}\right) dx = \frac{2}{\sqrt{\pi}} (k_B T)^{3/2}$$

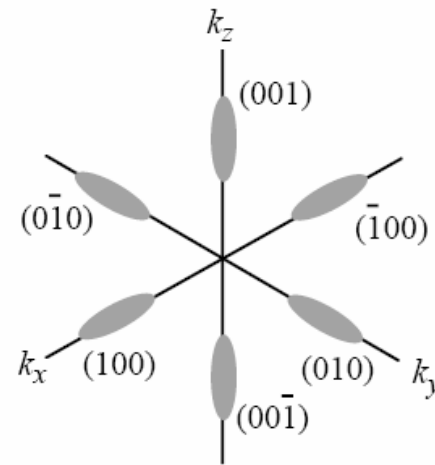
$$n = N_c(T) \exp\left(\frac{E_F - E_c}{k_B T}\right) = \frac{2D_c}{\sqrt{\pi}} (k_B T)^{3/2} \exp\left(\frac{E_F - E_c}{k_B T}\right)$$

$$N_c = \frac{2D_c}{\sqrt{\pi}} (k_B T)^{3/2} = 2 \left(\frac{m^* k_B T}{2\pi \hbar^2} \right)^{3/2} = \text{effective density of states}$$

Density of electrons in the conduction band



$$n = 2 \left(\frac{m^* k_B T}{2\pi \hbar^2} \right)^{3/2} \exp \left(\frac{E_F - E_c}{k_B T} \right)$$



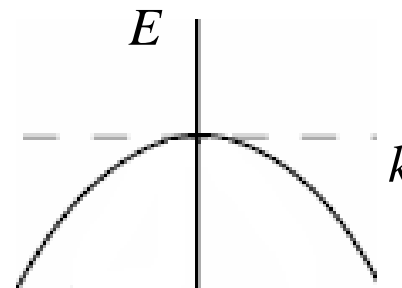
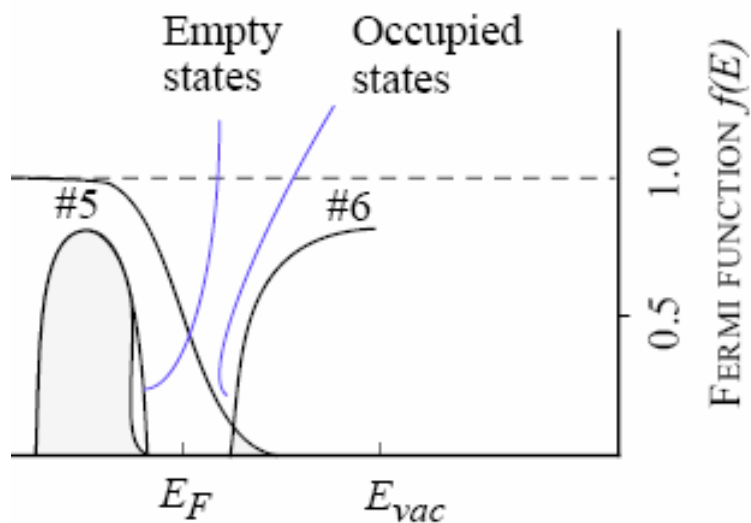
$$n = N_c \left(\frac{T}{300} \right)^{3/2} \exp \left(\frac{E_F - E_c}{k_B T} \right)$$

Properties	Si	Ge	GaAs
Bandgap E_g	1.12 eV	0.66 eV	1.424 eV
Effective density of states in conduction band (300 K) N_c	$2.78 \times 10^{25} \text{ m}^{-3}$	$1.04 \times 10^{25} \text{ m}^{-3}$	$4.45 \times 10^{23} \text{ m}^{-3}$

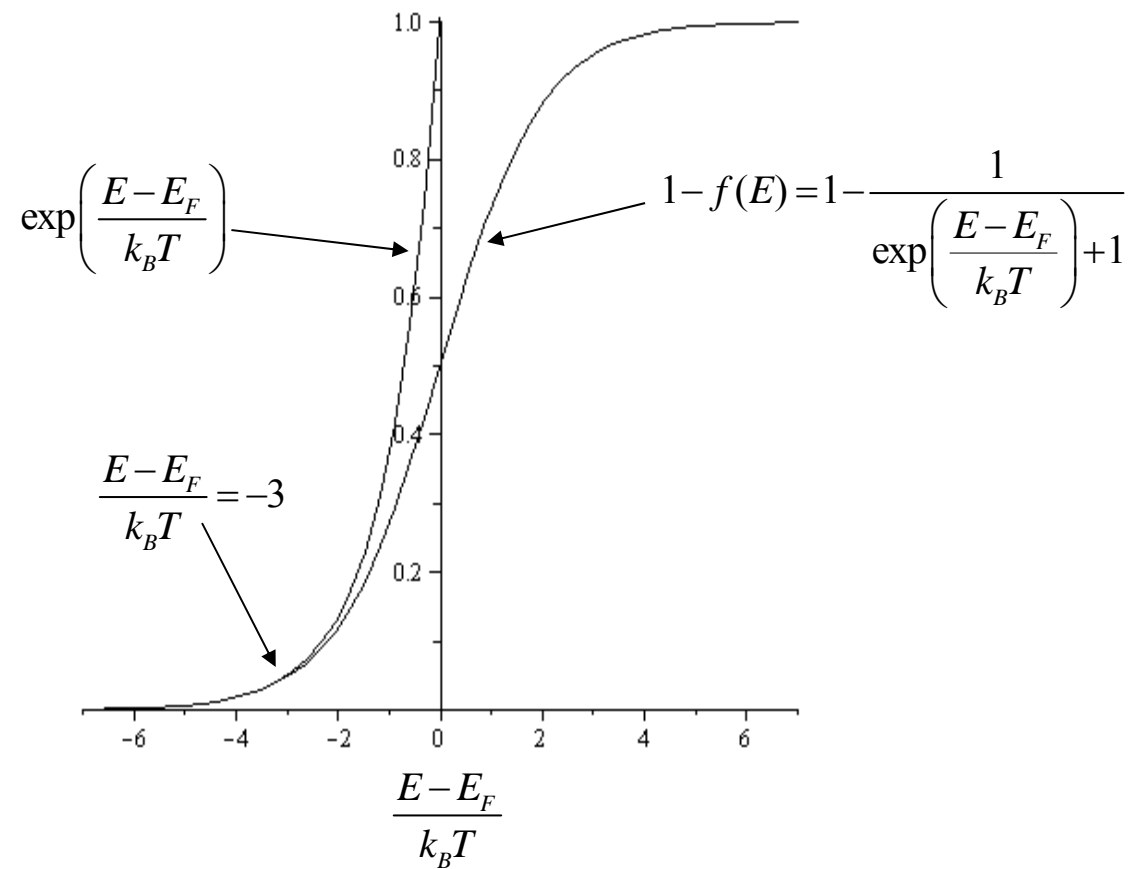
Density of holes in the valence band

$$D(E) = D_v \sqrt{E_v - E}$$

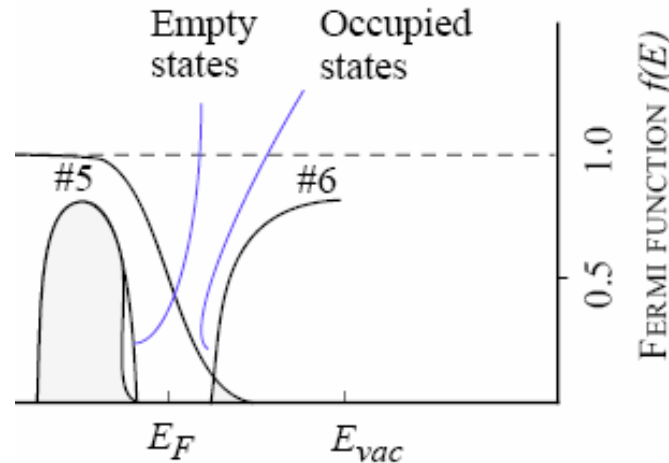
$$1 - f(E) = 1 - \frac{1}{1 + \exp\left(\frac{E - E_F}{k_B T}\right)} \approx \exp\left(\frac{E - E_F}{k_B T}\right)$$



Boltzmann approximation



Density of holes in the valence band

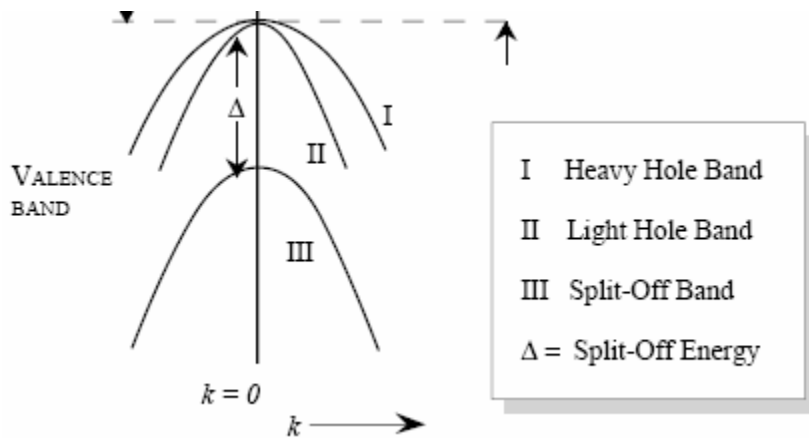


$$p = \int_{-\infty}^{E_v} D(E)(1 - f(E)) dE \approx D_v \int_{-\infty}^{E_v} \exp\left(\frac{E - E_F}{k_B T}\right) \sqrt{E_v - E} dE$$

$$p = N_v \exp\left(\frac{E_v - E_F}{k_B T}\right) = \frac{2D_v}{\sqrt{\pi}} (k_B T)^{3/2} \exp\left(\frac{E_v - E_F}{k_B T}\right)$$

$$N_v = 2 \left(\frac{m_h^* k_B T}{2\pi \hbar^2} \right)^{3/2} = \text{Effective density of states in the valence band}$$

Density of holes in the valence band



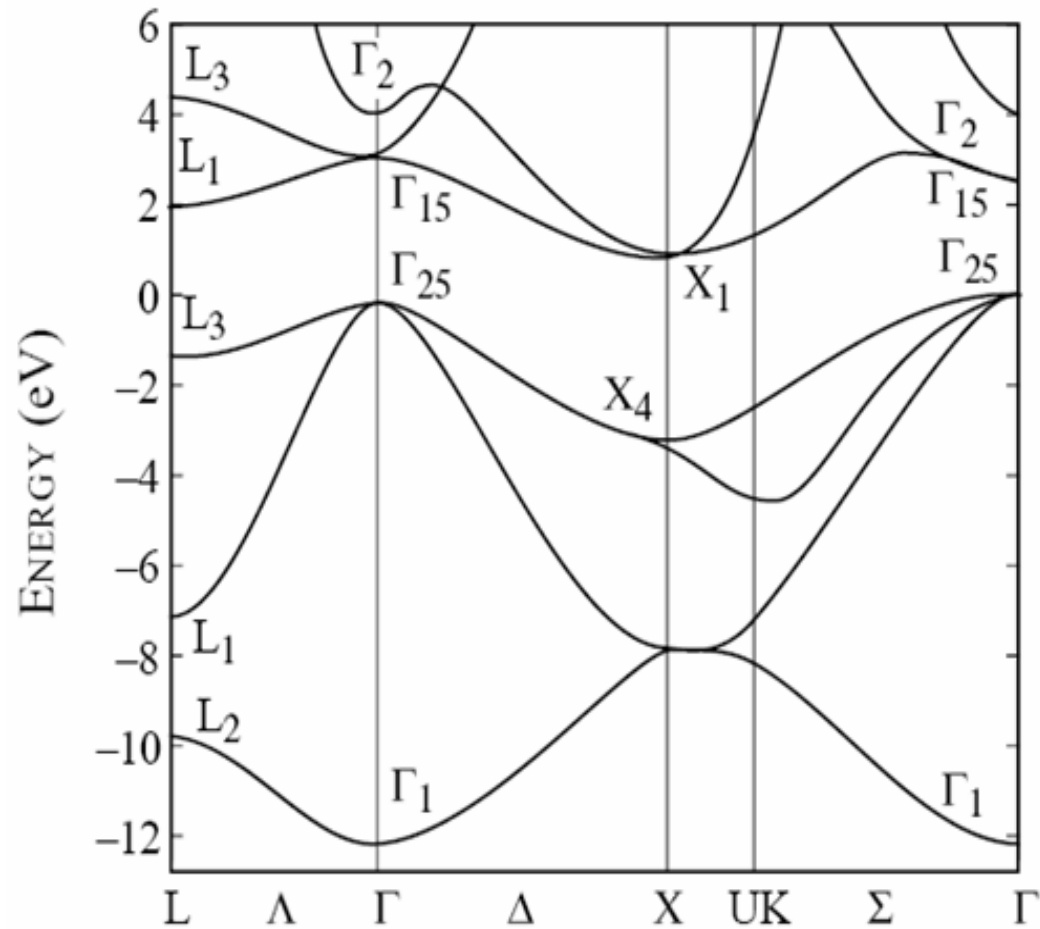
$$p = 2 \left(\frac{m_h^* k_B T}{2\pi \hbar^2} \right)^{3/2} \exp \left(\frac{E_v - E_F}{k_B T} \right)$$

$$p = N_v \left(\frac{T}{300} \right)^{3/2} \exp \left(\frac{E_v - E_F}{k_B T} \right)$$

Properties	Si	Ge	GaAs
Bandgap E_g	1.12 eV	0.66 eV	1.424 eV
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Effective density of states in valence band (300 K) N_v	$9.84 \times 10^{24} \text{ m}^{-3}$	$6.0 \times 10^{24} \text{ m}^{-3}$	$7.72 \times 10^{24} \text{ m}^{-3}$
Effective mass electrons m^*/m_0	$m_l^* = 0.98$ $m_t^* = 0.19$	$m_l^* = 1.64$ $m_t^* = 0.082$	$m^* = 0.067$
Effective mass holes m^*/m_0	$m_{lh}^* = 0.16$ $m_{hh}^* = 0.49$	$m_{lh}^* = 0.044$ $m_{hh}^* = 0.28$	$m_{lh}^* = 0.082$ $m_{hh}^* = 0.45$
Crystal structure	diamond	diamond	zincblende
Density	2.328 g/cm ³	5.3267 g/cm ³	5.32 g/cm ³
Atoms/m ³	5.0×10^{28}	4.42×10^{28}	4.42×10^{28}

Exam March 2007 Problem 1

The band structure of a semiconductor is shown below. The zero of energy is chosen to be the top of the valence band.

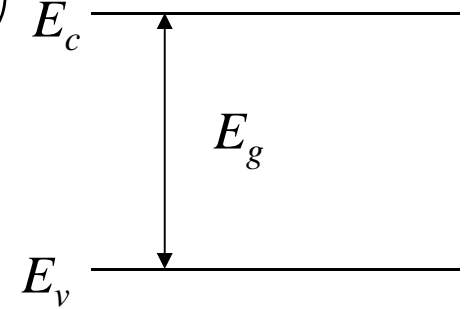


- Is this a direct or an indirect semiconductor? Why?
- What is the band gap?
- What are light holes and heavy holes? Explain how you can determine the effective mass of the holes from this diagram.

Law of mass action

$$np = N_c \exp\left(\frac{E_F - E_c}{k_B T}\right) N_v \exp\left(\frac{E_v - E_F}{k_B T}\right)$$

$$np = N_c N_v \exp\left(\frac{-E_g}{k_B T}\right)$$

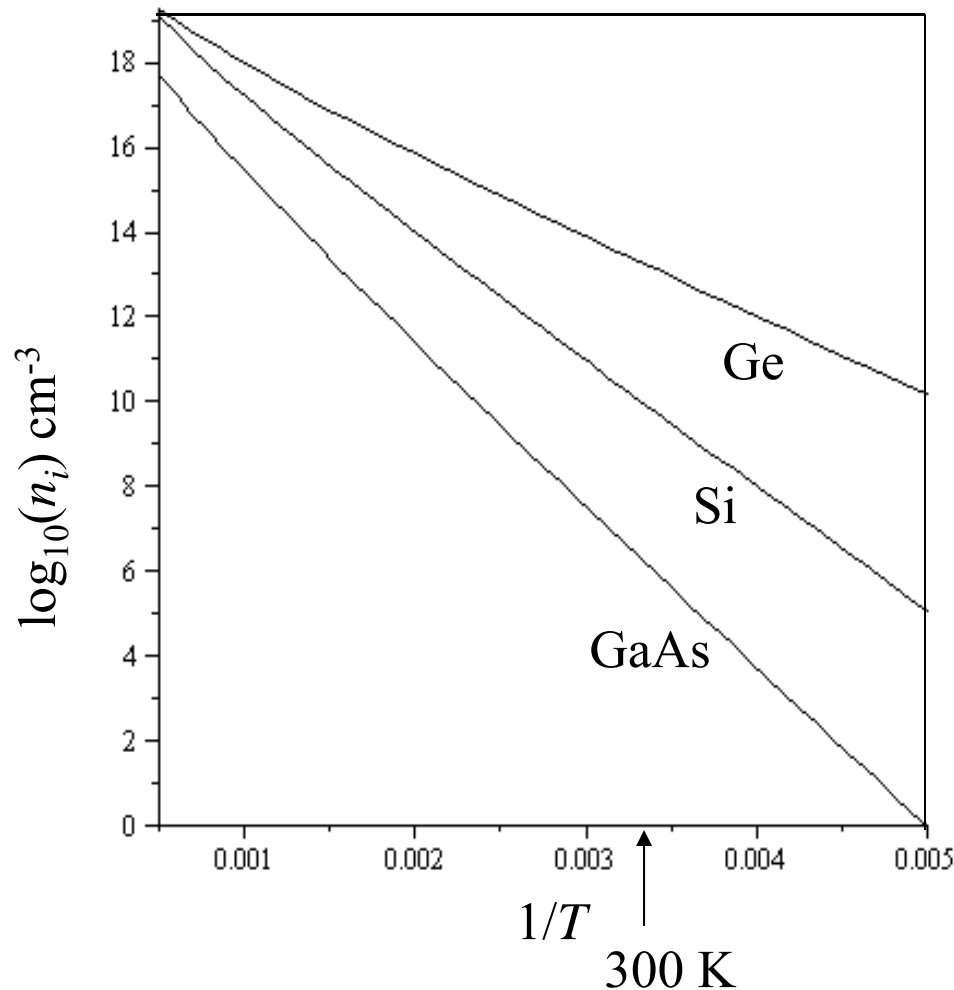


For intrinsic semiconductors (no impurities)

$$n = p = n_i = \sqrt{N_c N_v} \exp\left(\frac{-E_g}{2k_B T}\right)$$

intrinsic carrier density

Intrinsic carrier concentration



$$n_i = \sqrt{N_v N_c} \left(\frac{T}{300} \right)^3 \exp\left(-\frac{E_g}{2k_B T} \right)$$

Silicon has $\sim 5 \times 10^{22}$ atoms/ cm^3

Good for thermometer, bad for designing circuits.

Fermi energy of an intrinsic semiconductor

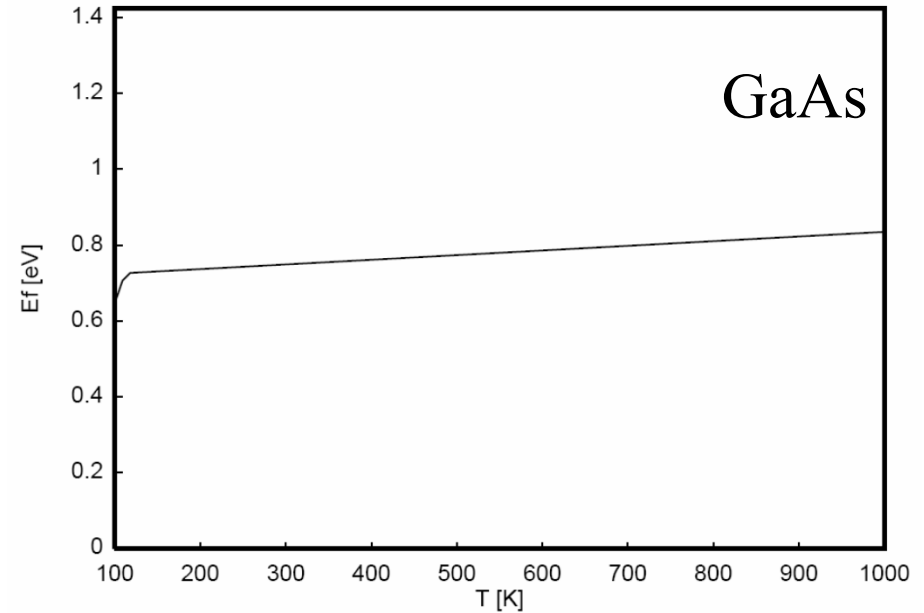
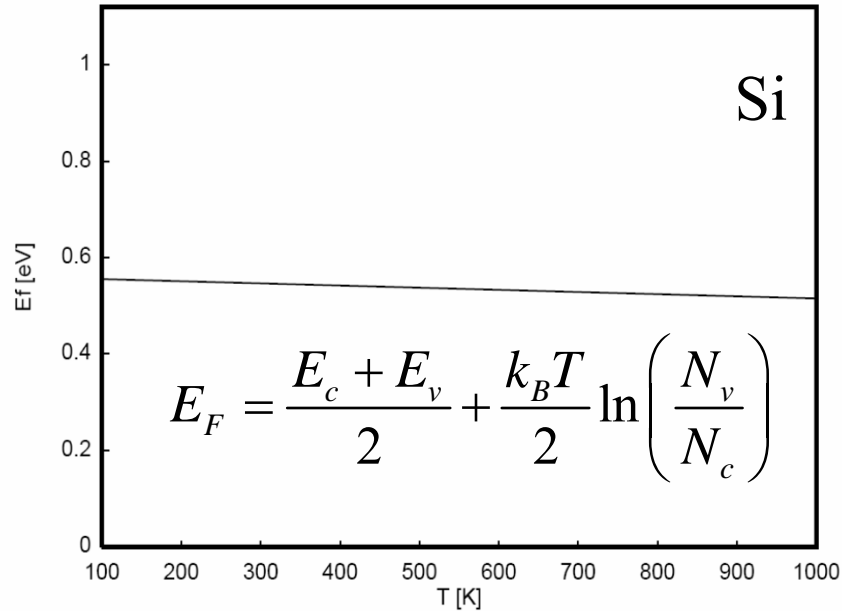
$$n = p = N_c \exp\left(\frac{E_F - E_c}{k_B T}\right) = N_v \exp\left(\frac{E_v - E_F}{k_B T}\right)$$

$$\frac{N_v}{N_c} = \exp\left(\frac{E_F - E_c - E_v + E_F}{k_B T}\right)$$

$$\frac{2E_F}{k_B T} = \frac{E_c + E_v}{k_B T} + \ln\left(\frac{N_v}{N_c}\right)$$

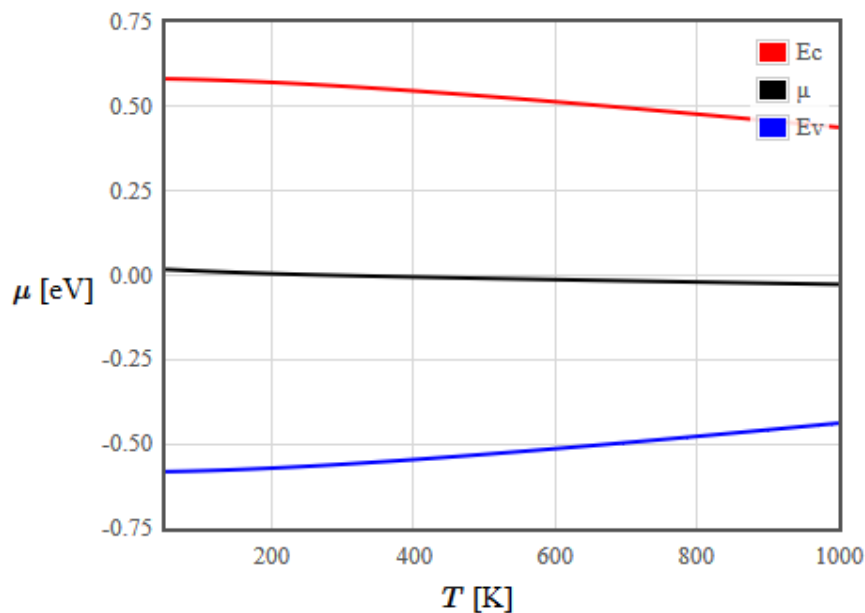
$$E_F = \frac{E_c + E_v}{2} + \frac{k_B T}{2} \ln\left(\frac{N_v}{N_c}\right)$$

Temperature dependence of E_F



Properties	Si	Ge	GaAs
Bandgap E_g	1.12 eV	0.66 eV	1.424 eV
Effective density of states in conduction band (300 K) N_c	$2.78 \times 10^{25} \text{ m}^{-3}$	$1.04 \times 10^{25} \text{ m}^{-3}$	$4.45 \times 10^{23} \text{ m}^{-3}$
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Intrinsic semiconductors with a split-off band



$N_c(300\text{ K}) =$	<input type="text" value="2.78E19"/>	1/cm ³	Semiconductor <input type="button" value="Si"/> <input type="button" value="Ge"/> <input type="button" value="GaAs"/>
$N_v(300\text{ K}) =$	<input type="text" value="9.84E18"/>	1/cm ³	
$N_{so}(300\text{ K}) =$	<input type="text" value="2.98E18"/>	1/cm ³	
$E_g =$	<input type="text" value="1.166-4.73E-4*T*T/(T+636)"/>	eV	
$E_v - E_{so} =$	<input type="text" value="0.044"/>	eV	
$T_1 =$	<input type="text" value="50"/>	K	
$T_2 =$	<input type="text" value="1000"/>	K	
<input type="button" value="Replot"/>			

http://lamp.tu-graz.ac.at/~hadley/ss1/semiconductors/intrinsic_so.php

Extrinsic semiconductors

The introduction of impurity atoms that can add electrons or holes is called doping.

n-type : donor atoms contribute electrons to the conduction band.
Examples: P, As in Si.

p-type : acceptor atoms contribute holes to the valence band.
Examples: B, Ga, Al in Si.

n and p

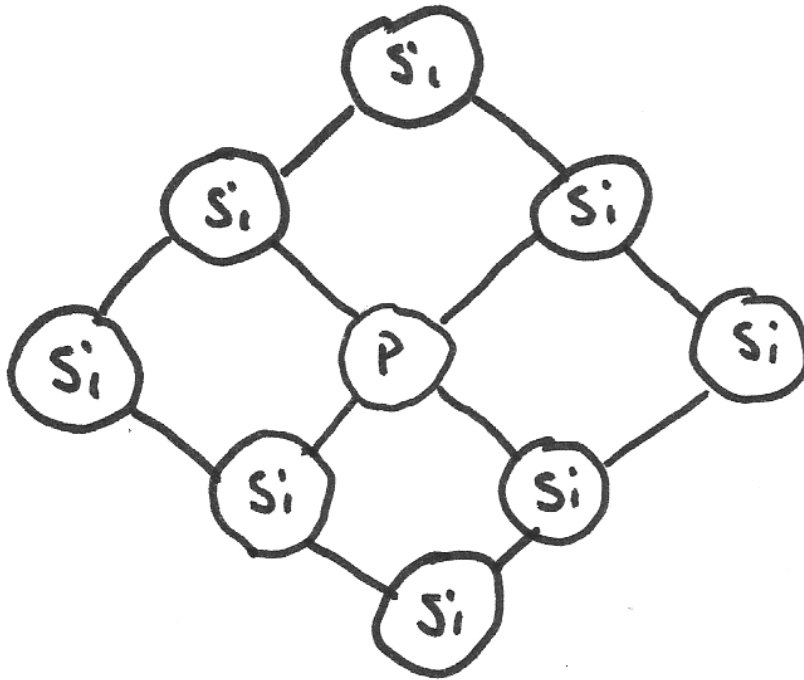
The electron density and hole density are:

$$n = N_c \exp\left(\frac{E_F - E_c}{k_B T}\right) \quad p = N_v \exp\left(\frac{E_v - E_F}{k_B T}\right)$$

The law of mass action:

$$np = n_i^2 = N_v N_c \exp\left(-\frac{E_g}{k_B T}\right)$$

Ionization of dopants



Easier to ionize a P atom in Si than a free P atom

$$E_n = -\frac{me^4}{8\epsilon_0^2 h^2 n^2}$$

Ionization energy is smaller by a factor:

$$\frac{m^*}{m} \left(\frac{\epsilon_0}{\epsilon_r \epsilon_0} \right)^2$$

Ionization energy ~ 25 meV

	IIIA	IVA	VA	VIA
	5 B	6 C	7 N	8 O
	13 Al	14 Si	15 P	16 S
IIIB	30 Zn	31 Ga	32 Ge	33 As
	34 Se			
	48 Cd	49 In	50 Sn	51 Sb
				52 Te

acceptors in Si

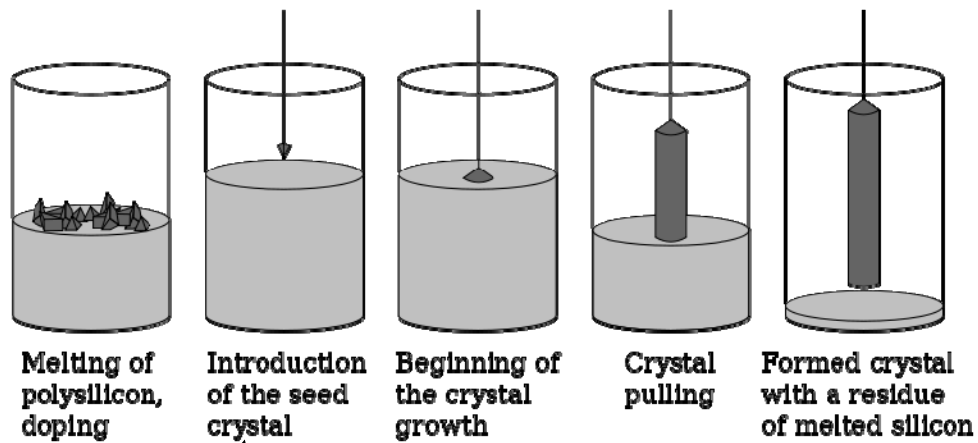


donors in Si



Crystal growth

Czochralski Process



add dopants to the melt



images from wikipedia

Crystal growth

Float zone Process

Neutron transmutation

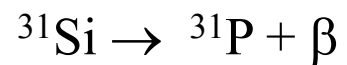
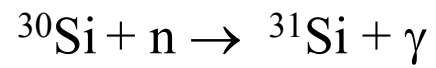


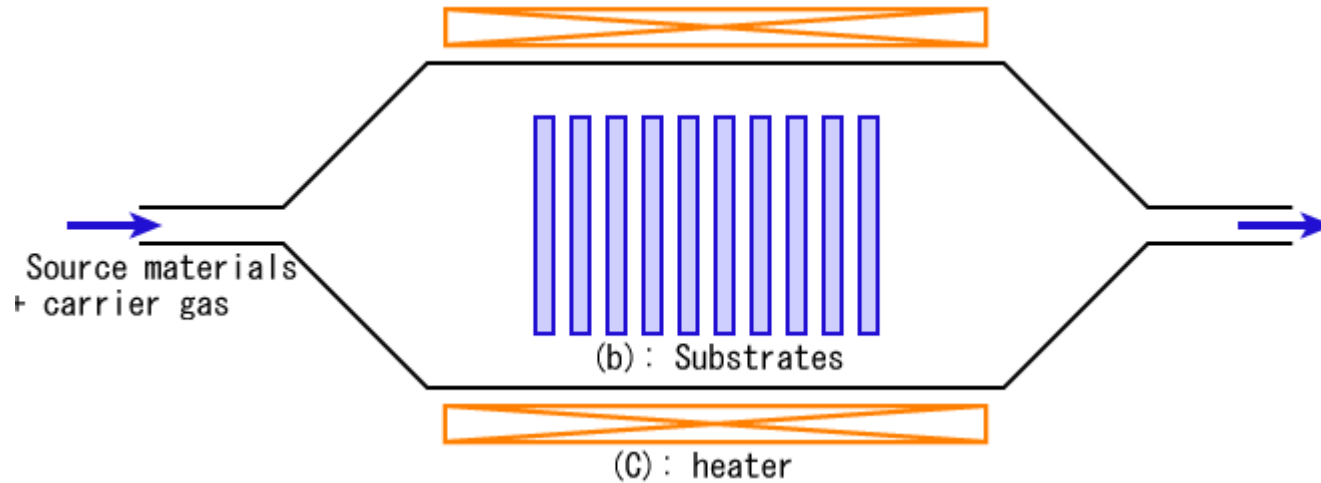
image from wikipedia

Gas phase diffusion



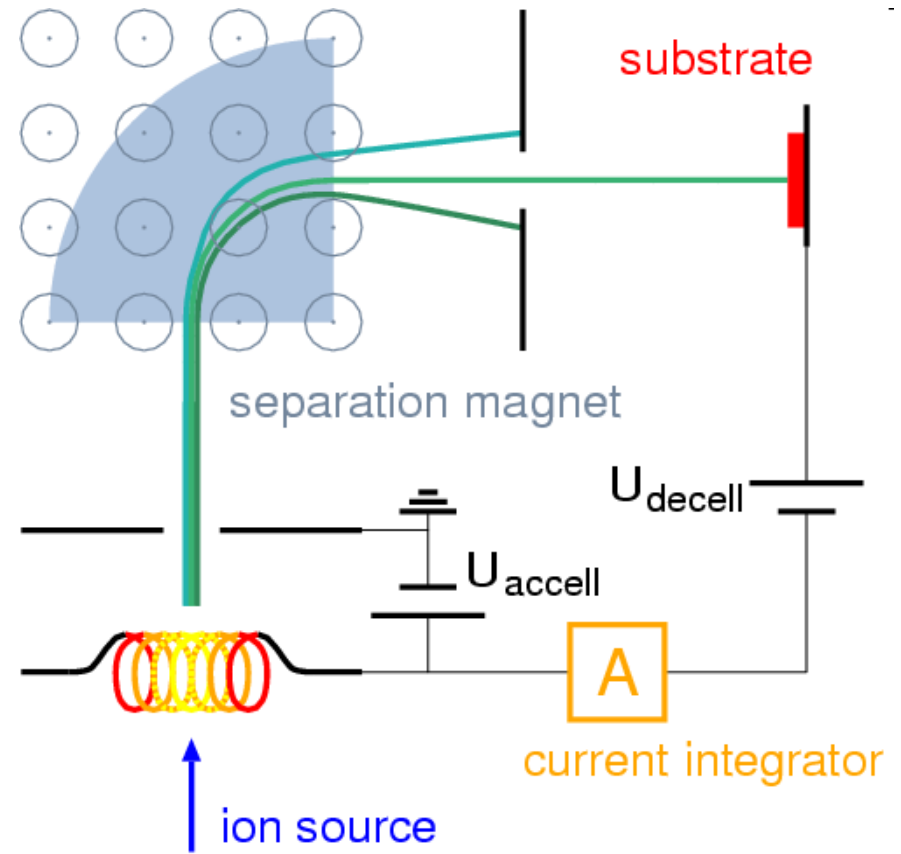
AsH_3 (Arsine) or PH_3 (phosphine) for n-doping
 B_2H_6 (diborane) for p-doping.

Chemical vapor deposition



Epitaxial silicon CVD SiH_4 (silane) or SiH_2Cl_2 (dichlorosilane)
 PH_3 (phosphine) for n-doping or B_2H_6 (diborane) for p-doping.

Ion implantation



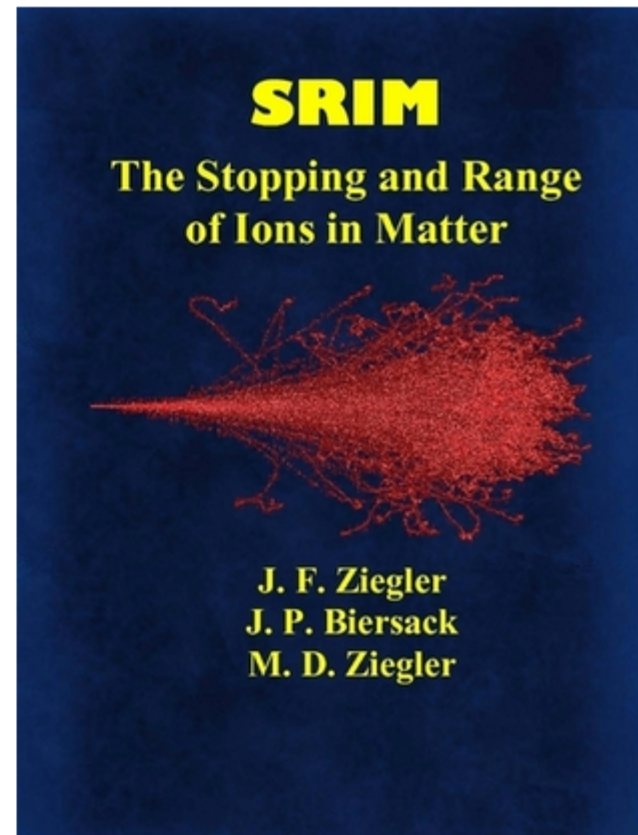
Implant at 7° to avoid channeling

SRIM

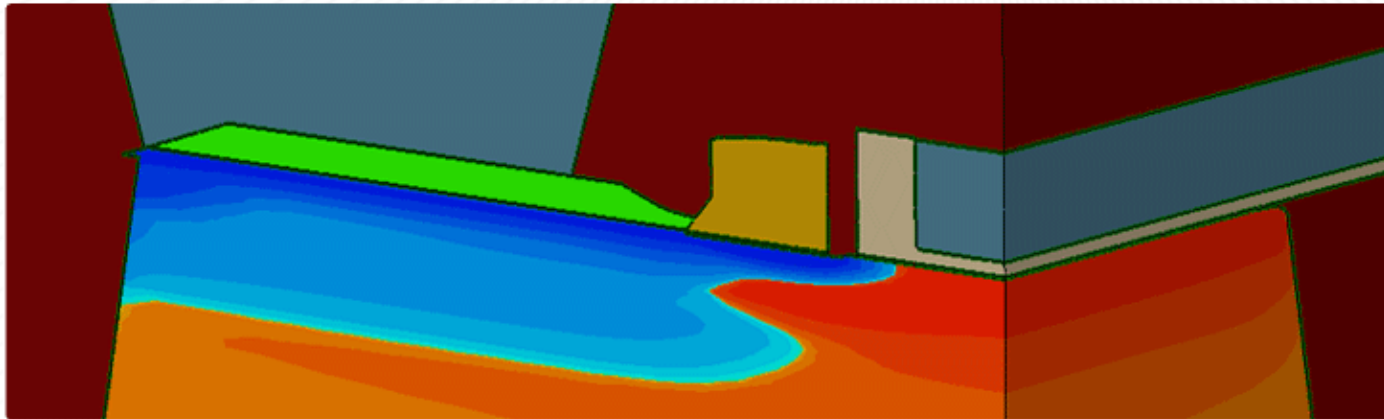
The Stopping and Range of Ions in Matter

James F. Ziegler, Jochen P. Biersack, Matthias D. Ziegler

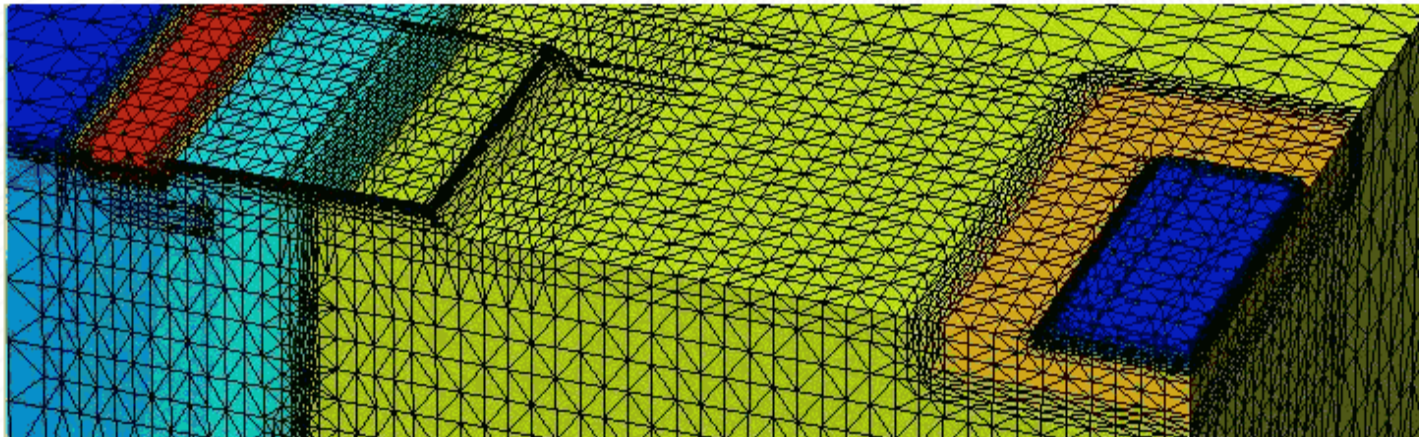
- Ch 1 - **Historical Review**
- Ch 2 - **Nuclear Stopping of Ions**
- Ch 3 - **Electronic Stopping of Ions**
- Ch 4 - **Stopping of Energetic Light Ions**
- Ch 5 - **Stopping of Ions in Compounds**
- Ch 6 - **Ion Straggling**
- Ch 7 - **TRIM : Scientific Background**
- Ch 8 - **TRIM : Setup and Input**
- Ch 9 - **TRIM : Output Files**
- Ch 10 - **Stopping and Range Tables**
- Ch 11 - **SRIM Tutorials**



Process Simulation



Device Simulation



<http://www.synopsys.com>